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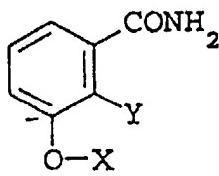
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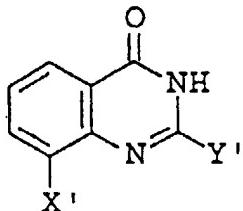
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(54) Title: BENZAMIDE ANALOGS, USEFUL AS PARP (ADP-RIBOSYLTRANSFERASE, ADPRT) DNA REPAIR ENZYME INHIBITORS



(I)



(II)

## (57) Abstract

A range of 3-oxybenzamide compounds (I) and related quinazolinone compounds (II) are disclosed which can act as potent inhibitors of the DNA repair enzyme poly(ADP-ribose) polymerase or PARP enzyme (EC 2.4.2.30), and which thereby can provide useful therapeutic compounds for use in conjunction with DNA-damaging cytotoxic drugs or radiotherapy to potentiate the effects of the latter. The compounds disclosed include 3-benzyloxybenzamides, 3-oxybenzamides in which a chain of 5 or more methylene groups terminate in a halogen atom or in a purin-9-yl moiety, certain benzoxazole-4-carboxamide compounds and certain quinazolinone compounds. In formula (I) X and Y together may form a bridge -X-Y- that represents the grouping (a), (b) or (c) wherein R<sup>5</sup> is H, alkyl, aryl or aralkyl.

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BENZAMIDE ANALOGS, USEFUL AS PARP (ADP-RIBOSYLTRANSFERASE, ADPRT) DNA REPAIR ENZYME INHIBITORS

The present invention relates to benzamide analogues, especially certain 3-substituted benzamide compounds and 5 related quinazolinone compounds that are of interest as being at least potentially useful chemotherapeutic agents by virtue of an ability to inhibit the activity of the enzyme poly ADP-ribosyltransferase (EC 2.4.2.30), also known as poly(ADP-ribose) polymerase, commonly referred to as ADPRT or PARP. In general, the latter abbreviation, PARP, will be used throughout the present specification.

BACKGROUND

At least in higher organisms, the enzyme poly ADP-ribosyltransferase is known to catalyse a transfer of the ADP-ribose moiety from the oxidized form NAD<sup>+</sup> of nicotinamide adenine dinucleotide to nuclear acceptor proteins so as to form homo ADP-ribose polymers, and this process has been implicated in a number of cellular events such as, for example, repair of DNA damage, development of cellular differentiation, transformation of cells by oncogenes, and gene expression. A common feature in a number of these processes is the formation and repair of DNA strand breaks and the stage which involves the PARP enzyme appears to be that of DNA ligase II-mediated strand rejoining. In the majority of cases a role for poly ADP-ribosylation has been implicated by the use of inhibitors of the PARP enzyme, and this has led to suggestions that such inhibitors, by interfering with the intracellular DNA repair mechanism, may have a useful chemotherapeutic role insofar as they should be able to modify treatment resistance characteristics and potentiate or enhance the effectiveness of cytotoxic drugs in chemotherapy or of radiation in radiotherapy where a primary effect of the treatment is that of causing DNA damage in target cells, as for example in many forms of antitumour therapy.

In this connection, several classes of PARP inhibitors are already known, including benzamide itself

and various nicotinamide and benzamide analogues, especially 3-substituted benzamides with small substituent groups such as 3-amino, 3-hydroxy and 3-methoxy. PARP inhibitory activity of certain N-substituted benzamides has 5 also been reported in EP-A-0305008 wherein it has also been proposed to use these compounds in medicine for increasing the cytotoxicity of radiation or of chemotherapeutic drugs.

10        Regarding this use of benzamides as chemotherapeutic agents, a number of studies on such compounds that are known to exhibit PARP inhibitory activity have confirmed that they can potentiate the cytotoxicity of a range of antitumour agents *in vitro*, for example, bleomycin and 15 methylating drugs. More limited data has further indicated that such benzamides can also potentiate the activity of cytotoxic drugs *in vivo*, although the dose requirements have appeared to be rather high (e.g. in the region of 0.5g kg<sup>-1</sup> per dose for 3-aminobenzamide) and there may be 20 associated problems in preparing satisfactory pharmaceutical formulations and in avoiding toxicity limitations. Furthermore, a number of the known benzamides have also been shown clearly to have potential as radiosensitizers, increasing for example ionising radiation-induced tumour 25 cell kill both *in vitro* and *in vivo*, and it is believed that in many cases this effect is related to these compounds acting as PARP inhibitors and interfering with DNA repair.

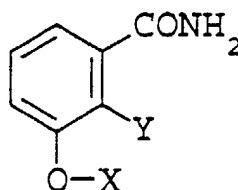
30        However, notwithstanding the existing data from *in vitro* and *in vivo* studies suggesting that PARP inhibitors have considerable potential as useful chemotherapeutic agents which merit further clinical evaluation, for 35 instance in connection with cancer therapy, currently available known PARP inhibitors are not considered as yet to be entirely suitable to represent candidate drugs. Accordingly, there is a need to find and develop a greater range of compounds having potentially useful PARP inhibitory properties.

DISCLOSURE OF THE INVENTION

The present invention identifies a new range or ranges of compounds of interest as PARP inhibitors that can be useful in medicine, especially when administered in conjunction with at least certain cytotoxic drugs or with radiotherapy for increasing the cytotoxic effectiveness thereof. In general, the compounds of this invention as hereinbelow defined comprise novel 3-substituted benzamide compounds, especially 3-oxybenzamide compounds, or analogues, of which many include relatively large or bulky 3-position substituents or include 3-position substituents linked in a ring structure with substituents in the 2-position. The compounds also include certain quinazolinones of which at least some may be formed by molecular rearrangement of related benzamide compounds. By virtue of their structure in general such compounds are adapted to act as an alternative substrate to NAD<sup>+</sup> for the PARP enzyme.

More specifically, from one aspect, the invention provides novel compounds selected from: (A), a 3-substituted benzamide compound having the general structural formula I

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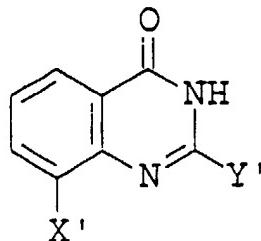


I

or a pharmaceutically acceptable salt thereof, and

(B), a quinazolinone compound having the general structural formula II

35



II

or a pharmaceutically acceptable salt thereof,

characterised in that in structural formula I

(i) Y is hydrogen, and

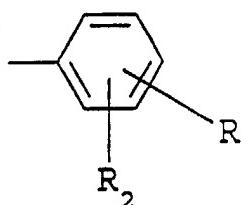
X is  $-\text{CH}_2\text{-Z}$

5 wherein

Z represents alkyl, an optionally substituted aralkyl group,  $-\text{CH}=\text{CHR}$  (where R is H, alkyl or an optionally substituted phenyl group), cyclohexyl, or a group having the structural formula III

10

15



III

where  $R_1$  is selected from H, alkoxy,  $\text{NO}_2$ ,  $\text{N}_3$ ,  $\text{NH}_2$ ,  $\text{NHCOR}_3$  ( $R_3$  being alkyl or aryl),  $\text{CO}_2\text{R}_4$  ( $R_4$  being H or alkyl), alkyl, hydroxyalkyl,  $\text{CW}_3$  or W (W being halide), and CN,

and where

25

$R_2$  is H,

or where  $R_1$  and  $R_2$  together represent a group  $-\text{O-CHR}_5\text{-O-}$  bridging adjacent ring C's with  $R_5$  being H, alkyl or an optionally substituted aralkyl or aryl group

30

or

(ii) Y is hydrogen, and

X is  $-(\text{CH}_2)_n\text{-Z}$

35

wherein n is in the range of 5 to 12, and

Z is halide or a purin-9-yl moiety;

or

(iii) Y and X together form a bridge  $-\text{Y-X-}$  that represents the grouping  $-\text{N=C-}$  or  $-\text{O-CH-}$  or  $-\text{S-CH-}$



where R<sub>5</sub> is as specified above,

and in structural formula II

5

X' represents hydroxyl, alkyl, alkoxy or an optionally substituted aryl (e.g. phenyl) or aralkyl (e.g. benzyl) group,

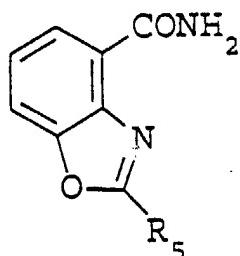
and

10 Y' represents hydrogen, alkyl or an optionally substituted aryl (e.g. phenyl) or aralkyl (e.g. benzyl) group.

Alkyl groups when present as such or as a moiety in  
15 other groups such as alkoxy, excluding in some cases the  
methylene chain -(CH<sub>2</sub>)<sub>n</sub>- specified above, will generally be  
composed of 1-8 carbon atoms, preferably 1-6 carbon atoms,  
and more usually 1-4 carbon atoms.

20 One very important group of compounds of special interest from the point of view of PARP-inhibitory activity comprises benzoxazole-4-carboxamide compounds, i.e. compounds represented by the formula IV

25



IV

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where R<sub>5</sub>, if not H, is preferably alkyl, phenyl or another aryl group such as naphthyl or pyridyl. When R<sub>5</sub> is an alkyl group this will generally be C<sub>1-6</sub> alkyl, such as for example methyl, ethyl, n-propyl, i-propyl, n-butyl, t-butyl or cyclohexyl. However, it may in some cases be larger, such as in adamantyl for instance. When R<sub>5</sub> is a phenyl group this may be substituted, especially in the 4 (para) position but alternatively perhaps in the 2-position or 3-position, by substituents such as alkoxy for example.

Within this group of benzoxazole compounds preferred members which are of particular interest include

- 5           2-methylbenzoxazole-4-carboxamide,  
          2-t-butylbenzoxazole-4-carboxamide,  
          2-phenylbenzoxazole-4-carboxamide,  
          2-(4-methoxyphenyl)benzoxazole-4-carboxamide.

10          In the above-mentioned compounds of formula IV, wherein there is an electron-rich aromatic ring, it is believed that the carboxamide group is constrained in a fixed conformation, particularly favourable for presenting the compound as an alternative substrate to NAD<sup>+</sup> for the  
15 PARP enzyme, by an intramolecular hydrogen bond between the ring nitrogen atom and one of the hydrogen atoms of the carboxamide group. A similar, although probably somewhat weaker effect may also occur in other compounds of formula I where the X and Y substituents form a bridge, as defined  
20 under (iii) above, containing an oxygen or sulphur atom, i.e. where the ring N atom of the benzoxazoles of formula IV is replaced by an O or S atom.

25          It has, however, also been found that in attempting to prepare benzoxazole-4-carboxamide compounds of formula IV, in some methods of preparation which could be expected to yield the desired compound the product is liable to undergo a molecular rearrangement (especially if liquid  
30 ammonia is used to form the carboxamide) and an 8-hydroxy quinazolinone derivative is obtained instead of the expected benzoxazole. Unexpectedly, it has been found that at least some such quinazolinone derivatives, which may of course be prepared by various methods, possess a  
35 potentially very useful biological activity as PARP inhibitors of high activity. Accordingly, these quinazolinone compounds which generally conform to structural formula II represent another very important aspect of the present invention. Examples of such compounds which are of particular interest include:

- 5                     (a) 8-hydroxy-2-methylquinazolin-4-[3H]one;  
                       (b) 8-hydroxyquinazolin-4-[3H]one;  
                       (c) 8-hydroxy-2-(4-nitrophenyl)-quinazolin-4-one;  
                       (d) 8-methoxy-2-methylquinazolin-4[3H]-one;  
                       (e) 8-methoxy-2-phenylquinazolin-4[3H]-one;  
                       (f) 8-hydroxy-2-phenylquinazolin-4[3H]-one;  
                       (g) 2,8-dimethylquinazolin-4[3H]-one.

10                  Another important group of compounds of particular interest comprises 3-benzyloxybenzamides (BOB benzamide analogues) where X is a benzyl or substituted benzyl group. Examples of benzyl group substituents include 2-nitro (or another 2-substituent), 4-CH<sub>3</sub>, 4-CO<sub>2</sub>H, 4-CO<sub>2</sub>CH<sub>3</sub>, 4-CONH<sub>2</sub>, 4-CN, 4-CH<sub>2</sub>OH, 4-NHCOPh, and within this group specific 15 compounds of particular interest include

- 20                  3-benzyloxybenzamide,  
                       3-(4-methoxybenzyloxy)benzamide,  
                       3-(4-nitrobenzyloxy)benzamide,  
                       3-(4-azidobenzyloxy)benzamide,  
                       3-(4-bromobenzyloxy)benzamide,  
                       3-(4-fluorobenzyloxy)benzamide,  
                       3-(4-aminobenzyloxy)benzamide,  
                       25         3-(3-nitrobenzyloxy)benzamide,  
                       3-(3,4-methylenedioxyphenylmethyloxy)benzamide or  
                       3-(piperonyloxy)benzamide,  
                       3-(N-acetyl-4-aminobenzyloxy)benzamide,  
                       3-(4-trifluoromethylbenzyloxy)benzamide,  
                       30         3-(4-cyanobenzyloxy)benzamide,  
                       3-(4-carboxymethylbenzyloxy)benzamide,  
                       3-(2-nitrobenzyloxy)benzamide,  
                       3-(4-carboxybenzyloxy)benzamide.

35                  In some cases the aromatic ring of the benzyl moiety in compounds of the above group of BOB analogues may be hydrogenated and still show some PARP inhibitory activity, one example of a compound in this category being 3-(cyclohexylmethyloxy)benzamide.

A further important group of compounds in accordance with the invention comprises the 3-oxybenzamides where there is a chain of 5 or more methylene groups terminating in a halogen atom, e.g. Br, or in a purin-9-yl moiety, especially adenine or 6-chloropurine. Specific compounds of interest within this group include

- 3-(5-bromopentyloxy)benzamide,  
3-(8-adenos-9-yloctyloxy)benzamide,  
10 3-[5-(6-chloropurin-9-yl)pentyloxy]benzamide,  
3-(5-adenos-9-ylpentylloxy)benzamide,  
3-[8-(6-chloropurin-9-yl)octyloxy]benzamide,  
3-[12-(6-chloropurin-9-yl)dodecyloxy]benzamide,  
3-(12-adenos-9-yl)dodecyloxy)benzamide.

15

In addition, however, particularly interesting compounds are provided when the 3-position oxy-substituent includes a double bond such as in an allyl group, for example 3-allyloxybenzamide, or a cinnamyl group, for 20 example 3-cinnamylloxybenzamide.

In another group of preferred compounds the 3-position oxy-substituent comprises an alkyl group having at least 4 carbon atoms. Typical examples include

25

- 3-butoxybenzamide,  
3-pentyloxybenzamide,  
3-hexyloxybenzamide,  
3-heptyloxybenzamide,  
30 3-octyloxybenzamide.

The invention also embraces or extends to methods of preparing compounds as hereinbefore defined (including intermediates in some cases) and to the therapeutic use of 35 such compounds. This includes their use for making medical or veterinary preparations or pharmaceutical formulations containing an effective PARP inhibitory amount of the active compound for administration to a patient in conjunction with a cytotoxic drug or radiotherapy in order

to increase the cytotoxic effectiveness of the latter. Such preparations or formulations may be made up in accordance with any of the methods well known in the art of pharmacy for administration in any suitable manner, for example orally, parenterally (including subcutaneously, intramuscularly or intravenously), or topically, the mode of administration, type of preparations or formulation and the dosage being generally determined by the details of the associated cytotoxic drug chemotherapy or radiotherapy that 10 is to be enhanced.

As indicated, the compounds according to this invention have at least potential as PARP inhibitors, and in vitro tests hereinafter described have demonstrated 15 positive pharmacological activity which it is believed reflects the activity to be found in vivo in the course of therapeutic clinical use.

It will be understood that where reference is made in 20 this specification to compounds of formula I or II (or formula IV) such reference should be construed as extending also to their pharmaceutically acceptable salts where relevant. Also, where any of the compounds referred to can exist in more than one enantiomeric form, all such forms, 25 mixtures thereof, and their preparation and uses are within the scope of the invention.

In general, many of the compounds of the present 30 invention, including at least the benzyloxybenzamide (BOB) and allyl 3-oxybenzamide analogues, are conveniently prepared by a base-catalysed alkylation of 3-hydroxybenzamide, e.g. reaction in the presence of acetonitrile and potassium carbonate, using an appropriate alkylating agent (e.g. an alkyl halide R-Hal) which, if not available 35 commercially, can be prepared via conventional methods. Initially, since 3-hydroxybenzamide itself is not widely available commercially, this compound can first be prepared by reacting commercially available 3-hydroxybenzoic acid with a mixture of triethylamine and ethyl chloroformate in

dichloromethane to give a mixed anhydride that is quenched in liquid ammonia, or by a newly developed efficient, high yield, preparative route involving selective acetylation of the 3-position -OH of 3-hydroxybenzoic acid and subsequent conversion of the carboxyl group to carboxamide, as hereinafter described.

10 DESCRIPTION OF EXAMPLES OF PREFERRED EMBODIMENTS

The following examples and descriptions of stages in synthetic routes of preparation of various preferred compounds of interest serve to further illustrate the present invention, but should not be construed in any way 15 as a limitation thereof.

In the first example (EXAMPLE 1), the above-mentioned new method of preparing 3-hydroxybenzamide from 3-20 hydroxybenzoic acid will be described since the 3-hydroxybenzamide is a starting material for the preparation of other benzamide analogues hereinafter described.

25 EXAMPLE 1

3-Hydroxybenzamide

(a) 1st Stage - Preparation of 3-Acetoxybenzoic acid

30 3-Hydroxybenzoic acid (1g; 7.29mM) was added to a cooled solution of sodium hydroxide (0.61g; 15.2mM dissolved in 2ml of water). Cooled acetic anhydride (0.81g; 7.9mM) was added with crushed ice (2g). The 35 mixture was stirred for 1 hour, and acidified with 6M hydrochloric acid (3ml) to yield a white precipitate. The organics were extracted into dichloromethane (3x30ml), and dried over magnesium sulphate. The solvent was filtered and then removed under vacuum to yield a white solid which was recrystallised from boiling water.

Solvent for Thin-layer chromatography (T.L.C.):

10% methanol/90% dichloromethane

NMR: 200 MHz:  $d_6$ DMSO:  $\delta$  = 2.4 (s; 3H; CH<sub>3</sub>); 7.45(m; 1H; H $\delta$ );  
7.6(t; 1H; H $\beta$ ); 7.78(m; 1H; H $\gamma$ ); 7.9(t; 1H; H $\alpha$ );

5 13.1(s; 1H; OH)

Yield 86%

(b) 2nd Stage - Preparation of 3-Hydroxybenzamide

10

3-Acetoxybenzoic acid (0.5g;) from the first stage was dissolved in thionyl chloride (1.74ml) and refluxed for 3½ hours.

15

Excess thionyl chloride was removed by distillation to yield a yellow oil, which was added dropwise to a cooled solution of ammonia (35% aq) and stirred for 30 minutes. The mixture was boiled to a reduced volume (50%) and left to cool. 3-hydroxybenzamide crystallised out of solution 20 and was collected and then recrystallised from boiling water.

T.L.C. : 10% methanol/90% dichloromethane

NMR: 200MHz;  $d_6$ DMSO;  $\delta$ : 7.0 (1H; dt; H<sub>4</sub>); 7.25(m; 4H; NH;  
H<sub>2</sub>; H<sub>5</sub>; H<sub>6</sub>); 8.0(s; 1H; NH); 9.77(s; 1H; OH).

25 Yield 60%

EXAMPLE 2

30

3-(4-Azidobenzyloxy)benzamide (Compound NU1013)

(a) 1st Stage - Preparation of p-Azidotoluene

35

p-Toluidine (9.3mM) was dissolved in hydrochloric acid (5M; 10ml), and the mixture cooled to <0°C. Sodium nitrite (10.28mM) was dissolved in a minimum amount of water, and this solution was added dropwise to the reaction over 30 minutes. The solution was stirred for 20 minutes,

then the presence of the oxidising agent was tested for using starch/iodide paper. Sodium azide (37.38mM) was then added slowly to the reaction mixture over a period of 1 hour (due to vigorous effervescing). The reaction was quenched, once all the sodium azide had been added, in water (50ml), and quantitatively transferred to a conical flask with water (50ml). The mixture was neutralised with sodium carbonate, until pH=6. The organics were extracted into dichloromethane (3 x 30ml), washed with water (2 x 10 20ml), dried over magnesium sulphate, and excess solvent removed under vacuum. The p-azidotoluene thus obtained was an oil and was isolated via chromatography (100% petroleum/ether 40-60).

<sup>1</sup>H NMR: d<sub>6</sub> DMSO: 6: 2.37 (s; 3H; CH<sub>3</sub>); 7.0 (d; 2H; H<sub>3</sub>; H<sub>5</sub>);  
15 7.3 (d; 2H; H<sub>2</sub>; H<sub>6</sub>)

(b) 2nd Stage - Bromination of p-Azidotoluene

To p-azidotoluene (2.2mM) prepared as above was added 20 N-bromosuccinimide (2.48mM), and azo-isobutyronitrile (0.214mM) in anhydrous benzene (5ml). This was left to reflux for 5 hours, with the reaction being monitored by TLC. The organics were extracted into diethyl ether (3 x 15ml), and water, dried over magnesium sulphate, and the excess solvent removed under vacuum. The p-azidobenzyl bromide formed was isolated using chromatographic techniques.

(c) Final Stage - Preparation of p-Azidobenzyl-3-Oxybenzamide

To 3-hydroxybenzamide (2.0mM) under nitrogen was added anhydrous acetonitrile (20ml), potassium carbonate (2.0mM), and p-azidobenzyl bromide (2.0mM). The mixture was refluxed for 3 hours, and the progress of the reaction monitored by TLC. The excess solvent was removed under vacuum, until dry, and the title compound was recrystallised from hot dichloromethane (minimum amount). A pale yellow crystalline solid was isolated and dried.

Melting point: 167-168°C

Infrared data:  $\text{cm}^{-1}$ : 3341; 3155; 2121; 2094; 1631; 1583.

Mass spectra:  $m/z$ : 269 ( $M^{+1}$ ) 252; 223; 167; 104 (100%); 93; 77.

5  $^1\text{H}$  NMR:  $d_6$ -DMSO  $\delta$ = 5.5 (s; 2H;  $\text{CH}_2$ ); 7.27 (m; 3H;  $\text{H}_2'$ ;  $\text{H}_6'$ ;  $\text{H}_4$ ); 7.41-7.6 (m; 6H;  $\text{H}_3'$ ;  $\text{H}_5'$ ;  $\text{H}_2$ ;  $\text{H}_6$ ;  $\text{H}_5$ ; NH); 8.0 (s; 1H; NH)

$^{13}\text{C}$  NMR:  $\delta$ = 69.098; 114.049; 118.164; 119.528; 120.359; 129.735; 129.886; 134.188; 136.096; 139.332; 158.480; 10 167.889

Elemental Analysis:

Expect C: 62.68%; H: 4.47%; N: 20.89%

Found C: 62.18%; H: 4.30%; N: 20.70%

15 EXAMPLE 3

3-(4-Bromobenzyl)benzamide (Compound NU1014)

To 3-hydroxybenzamide (2.0mM) under nitrogen was 20 added anhydrous acetonitrile (20ml), potassium carbonate (2.0mM), and p-bromobenzyl bromide (2.0mM). The mixture was refluxed for 3 hours, and the progress of the reaction monitored by TLC. The excess solvent was removed under vacuum, until dry, and the title compound was 25 recrystallised from hot dichloromethane (minimum amount). A white crystalline solid was isolated and dried.

Melting point: 160-161°C

Infrared data:  $\text{cm}^{-1}$ : KBr disc: 3323; 3146; 1670; 1622.

30  $^1\text{H}$  NMR:  $d_6$ DMSO:  $\delta$ =5.26 (s; 2H;  $\text{CH}_2$ ); 7.3 (dd; 1H;  $\text{H}_4$ ); 7.49-7.75 (m 9H; PhBr;  $\text{NH}_2$ ; PhO)

$^{13}\text{C}$  NMR:  $d_6$ DMSO:  $\delta$ =114.031; 118.123; 120.402; 121.295; 129.704; 130.107; 131.690; 136.094; 136.723; 158.371; 167.843.

35 Mass spectra:  $m/z$ : EI: 307 ( $M^{+1}$ ); 262; 212; 169 (100%); 101; 90.

Elemental analysis:

Expect C: 55.26%; H: 3.29%; N: 4.61%

Found C: 54.61%; H: 3.66%; N: 4.47%.

EXAMPLE 43-(4-Fluorobenzyl)benzamide (Compound NU1015)

5 To 3-hydroxybenzamide (2.0mM) under nitrogen was added anhydrous acetonitrile (30ml), potassium carbonate (2.0mM), and p-fluorobenzyl bromide (2.0mM). The mixture was refluxed for 14 hours, and the progress of the reaction monitored by TLC. The excess solvent was removed under  
10 vacuum, until dry, and the title compound was recrystallised from hot dichloromethane (minimum amount), and petrol ether 40/60. A white crystalline solid was isolated and dried.

Melting point: 161-162°C

15 Infrared data: KBr disc:  $\text{cm}^{-1}$ : 3366; 3171;  
 $^1\text{H}$  NMR:  $d_6$ DMSO:  $\delta$ =5.23 (s; 2H;  $\text{CH}_2$ ); 7.26-7.68 (m; 9H; NH; aromatics); 8.11 (s; 1H;  $\text{H}_4$ )

20  $^{13}\text{C}$  NMR:  $d_6$ DMSO:  $\delta$ = 68.914; 113.951; 115.405; 115.821; 118.124; 120.329; 129.329; 130.255; 130.421; 133.487; 136.056; 158.464; 159.692; 164.546; 167.853.

Elemental analysis:

Expect: C: 67.2; H: 4.8; N: 5.6  
25 Found: C: 67.79; H: 4.81; N: 5.65

EXAMPLE 53-(3-Nitrobenzyl)benzamide (Compound NU1017)

30 To 3-hydroxybenzamide (2.0mM) under nitrogen was added anhydrous acetonitrile (20ml), potassium carbonate (2.0mM), and m-nitrobenzyl bromide (2.0mM). The mixture was refluxed for 3 hours, and the progress of the reaction monitored by TLC. The excess solvent was removed under vacuum, until dry, and the title compound was recrystallised from hot dichloromethane (minimum amount). A white crystalline solid was isolated and dried.

Melting point: 162-163°C

Infrared data: KBr disc: 3362; 3171; 1655; 1622.

Mass spectrum: EI: m/z: 272(M+); 136(100%); 105; 90; 77.

5

$^1\text{H}$  NMR:  $d_6$ DMSO:  $\delta$ =5.42(s; 2H;  $\text{CH}_2$ ): 7.28-7.33  
(m; 1H;  $\text{H}\alpha$ ); 7.45-7.66 (m; 4H; NH;  $\text{H}_2$ ;  $\text{H}_5$ ;  $\text{H}_6$ ); 7.8  
(t; 1H;  $\text{H}_5$ ,); 8.0(m:2H; NH;  $\text{H}_6$ ,); 8.3 (m:1H;  $\text{H}_4$ ,): 8.4  
(m; 1H;  $\text{H}_2$ ,)

10  $^{13}\text{C}$  NMR:  $d_6$ DMSO:  $\delta$ =68.354; 114.041; 118.124;  
120.589; 122.317; 123.120; 129.812; 130.435; 134.406;  
136.132; 139.661; 148.185; 158.185; 167.774.

Elemental analysis:

15       Expect: C: 61.76; H: 4.41; N: 10.11  
Found : C: 61.53; H: 4.32; N: 10.01

20 EXAMPLE 6

3-(N-acetyl-4-aminobenzyl)benzamide (Compound NU1030)

25       To Compound NU1013 (100mg; 0.375mM) from Example 2  
was added thioacetic acid (2ml) and this was left to stir  
at room temperature until the reaction was completed. The  
reaction was followed by T.L.C. 10% MeOH: 90%  $\text{CH}_2\text{Cl}_2$ . The  
product was then isolated via chromatography, and  
recrystallised from ethyl acetate and ether, to yield a  
30 white crystalline solid (30%).

Melting point: 198-199°C

35        $^1\text{H}$  NMR:  $\delta$ : 2.19 (s; 3H;  $\text{CH}_3$ ); 5.2(s; 2H;  $\text{CH}_2$ ); 7.3(m; 1H;  $\text{H}_4$ );  
7.4-7.8 (m; 8H; NH;  $\text{H}_2$ ;  $\text{H}_5$ ;  $\text{H}_6$ ;  $\text{H}_2'$ ;  $\text{H}_3'$ ;  $\text{H}_4'$ ;  $\text{H}_6'$ )  
8.1 (s; 1H; NH); 10.1 (s; 1H; HNCO)

$^{13}\text{C}$  NMR:  $\delta$ : 29.223, 74.412, 118.899, 123.013, 124.087,  
125.108, 133.699, 134.557, 136.463, 140.921, 144.296,  
163.685, 172.781.

EXAMPLE 73-(Piperonyloxy)benzamide (Compound NU1020)5 1st Stage - Preparation of Piperonyl Chloride

Piperonyl alcohol (1g; 6.57mM) was dissolved in diethyl ether (10ml), and to this was slowly added concentrated hydrochloric acid (3.81ml). This was left to stir for 30 minutes. The excess solvent was removed to yield a colourless oil which crystallised at 4°C, (99%).

<sup>1</sup>H NMR: d: 4.8 (s,2H,CH<sub>2</sub>); 6.2(s,2H,OCH<sub>2</sub>O); 7.0(m,3H,aromatics)

15 2nd Stage - Preparation of Piperonyl BOB

To 3-hydroxybenzamide (137mg; 1mM) was added potassium carbonate (138mg; 1mM) and this was dissolved in anhydrous acetonitrile under a nitrogen atmosphere. To this was added piperonyl chloride (270mg; 1mM), and the mixture left to reflux for 15 hours. The organics were extracted into dichloromethane (3 x 20ml), dried over magnesium sulphate, and the solvent removed under vacuum. The product was isolated using chromatographic techniques, with 20% petrol:80% ethyl acetate as eluant. A white solid was isolated which was recrystallised from hot water (97%).

Melting point: 141-142°C

<sup>1</sup>H NMR: d<sub>6</sub>DMSO: δ: 5.1 (s;2H;CH<sub>2</sub>); 6.1(s,2H; OCH<sub>2</sub>O); 7.1 (m; 3H; H<sub>2'</sub>; H<sub>5'</sub>; H<sub>6'</sub>); 7.25 (dd; 1H; H<sub>4</sub>); 7.5 (m; 4H; NH; N<sub>2</sub>; H<sub>5</sub>; H<sub>6</sub>); 8.05 (s; 1H; NH)

<sup>13</sup>C NMR: d<sub>6</sub>DMSO: δ: 69.02; 101.36; 108.40; 108.80; 114.01; 118.12; 120.20; 121.89; 129.65; 136.00; 147.20; 146.62; 158.53; 167.84.

Mass spectrum: EI: 271 (M<sup>+</sup>); 135 (100%)

Elemental analysis:

Expected C: 66.42; H: 4.79; N: 5.16

Found C: 66.18; H: 4.47; N: 4.94

EXAMPLE 83-(4-Trifluoromethylbenzyloxy)benzamide (Compound NU1036)

5       3-hydroxybenzamide (0.137g; 1mmol) was dissolved in anhydrous acetonitrile (20ml) under a nitrogen atmosphere. To this was added potassium carbonate (0.138g; 1mmol) and 4-(trifluoromethyl)benzyl bromide (0.155ml; 1mmol). This mixture was left to reflux for 17 hours.

10

The reaction was followed by TLC. Upon completion the acetonitrile was removed under reduced pressure to leave a white solid which was dissolved in water. The organics were extracted into dichloromethane (3 x 30ml), 15 dried over magnesium sulphate, filtered and the solvent removed under reduced pressure to leave a white crystalline solid which was recrystallised from boiling ethyl acetate and petrol (60-80).

20 M/Z (EI): 295 (35%; M<sup>+</sup>) 159 (100%)

<sup>1</sup>H: 200MHz: d<sub>6</sub> DMSO: δ : 5.4 (2H; s; CH<sub>3</sub>); 7.3 (1H; m; H<sub>4</sub>); 7.6 (4H; m; NH; H<sub>2</sub>; H<sub>5</sub>; H<sub>6</sub>); 7.8 (2H; d; H<sub>2</sub>'); H<sub>6</sub>'); 7.9 (2H; d; H<sub>3</sub>'; H<sub>5</sub>'); 8.1 (1H; s; NH)

25      <sup>13</sup>C: 68.726; 114.011; 118.125; 120.513; 125.627; 125.693; 128.305; 129.007; 129.774; 136.133; 142.136; 158.296; 167.852

## Elemental Analysis:

Expected C: 61.02, H: 4.06, N: 4.75.

30      Found     C: 61.05/60.91, H: 3.94/3.96, N: 5.06/ 4.86.

EXAMPLE 93-(4-Cyanobenzyloxy)benzamide (Compound NU1037)

35      3-hydroxybenzamide (0.137g; 1mmol) was dissolved in anhydrous acetonitrile (20ml) under a nitrogen atmosphere. To this was added potassium carbonate (0.138g; 1mmol) and

4-cyanobenzyl bromide (0.138g; 1mmol). This was left to reflux for 5 hours.

The reaction was followed by TLC. Upon completion 5 the acetonitrile was removed under reduced pressure. Water was added to the remaining solid. The organics were extracted into dichloromethane (3 x 30ml), dried over magnesium sulphate, filtered and the solvent removed under reduced pressure to 1e a white crystalline solid which 10 was recrystallised from boiling ethyl acetate and petrol (68%).

M/Z (EI) 252 (18% M<sup>+</sup>); 153; 116 (75%);

IR cm<sup>-1</sup> 3362, 3179 (amide NH<sub>2</sub>); 2228 (C≡N)

15 <sup>1</sup>H: 200MHz: d<sub>6</sub> DMSO: δ : 5.4 (2H; s; CH<sub>2</sub>); 7.3 (1H; m; H<sub>4</sub>); 7.65 (6H; m; H<sub>2</sub>; H<sub>5</sub>; H<sub>6</sub>; NH; H<sub>2'</sub>; H<sub>6'</sub>); 8.0 (2H; d; H<sub>3'</sub>; H<sub>5'</sub>).

13C: 68.685; 110.848; 113.999; 118.153; 119.089; 120.555; 128.385; 129.3792; 132.775; 136.133; 143.080; 158.222;

20 167.815.

Elemental Analysis:

Expected C: 71.43, H: 4.76, N: 11.11.

Found: C: 71.27, H: 4.96, N: 10.67.

25

EXAMPLE 10

3-(4-carboxymethylbenzyloxy)benzamide (Compound NU1041)

30 To 3-hydroxybenzamide (1.37g; 10mmol) was added potassium carbonate (1.38g; 10mmol) and methyl 4-chloromethylbenzoate (1.84g; 10mmol) in anhydrous acetonitrile (50ml). The mixture was heated under reflux for 5 hours.

35 The acetonitrile was removed under reduced pressure, and the solids dissolved in water. The organics were extracted into dichloromethane (3 x 50ml) and pooled. The solvent was dried over magnesium sulfate and removed under reduced pressure.

The solid was recrystallised from boiling ethyl acetate and petrol, collected and dried (65%).

IR : cm<sup>-1</sup>, 3331; 3154, 2959. M/Z; 285 (48%; M<sup>+</sup>), 254 5 (38%); 149 (100%) 121 (75%).

<sup>1</sup>H: d<sub>6</sub> DMSO: δ = 3.9; (3H; s; CH<sub>3</sub>), 5.25 (2H; OCH<sub>2</sub>Ph), 7.25 (1H; d, H<sub>4</sub>), 7.4 (5H; m; NH<sub>2</sub>; H<sub>2</sub>; H<sub>5</sub>; -H<sub>6</sub>), 7.6 (2H; d; H<sub>2</sub>'; H<sub>6</sub>'), 709 (2H; d; H<sub>3</sub>'; H<sub>5</sub>').

<sup>13</sup>C: 52.41; 68.948. 114.016; 118.093; 120.459; 127.792; 10 129.299; 129.669; 136.097; 142.786; 158.33; 166.312; 167.822.

Elemental Analysis:

Expected C 67.37%; H 5.26% N 4.91%.  
15 Found: C 67.19%; H 5.16%; N 4.78%

EXAMPLE 11

20 3-(2-Nitrobenzyloxy)benzamide (Compound NU1042)

To 3-hydroxybenzamide (0.137g; 1mmol) was added potassium carbonate (0.138g; 1mmol) and 2-nitrobenzyl bromide (0.216g; 1mmol) in anhydrous acetonitrile (10ml).

25 The mixture was heated under reflux for 5 hours.

The acetonitrile was removed under reduced pressure, and the solids dissolved in water. The organics were extracted into dichloromethane (3 x 20ml) and pooled. The 30 solvent was dried over magnesium sulfate and removed under reduced pressure.

The solid was recrystallised from boiling ethyl acetate and petrol, collected and dried.  
35

IR cm<sup>-1</sup>: 3368; 3196, M/Z; 272 (3.8%; M<sup>+</sup>); 248; 217; 196; 181; 136 (100%).

<sup>1</sup>H: d<sub>6</sub> DMSO: δ = 5.5 (2H; OCH<sub>2</sub>Ph), 7.14 (1H; d, H<sub>4</sub>), 7.4 (4H; m; NH; H<sub>2</sub>'; H<sub>5</sub>'; H<sub>6</sub>), 7.26 (2H; m; H<sub>5</sub>'; H<sub>4</sub>'),

20

8.0 (1H; br; NH); 8.02 (1H; d; H<sub>3'</sub>).  
13C: 66.783; 113.854; 118.172; 120.753; 125.183; 129.490;  
129.549; 132.679; 132.339; 136.136.; 147.786;  
158.112; 167.778.

5

## Elemental Analysis:

Required C 61.76%; H 4.41% N 10.29%;  
Found C 61.49%; H 4.42%; N 10.11%.

10

EXAMPLE 123-(4-carboxybenzyloxy)benzamide (Compound NU1052)

15 To 3-(4-carboxymethylbenzyloxy)benzamide (Compound NU1041) (0.03g; 0.1mmol) was added methanol (3ml) and aqueous sodium hydroxide (1M; 3ml). This was warmed to 40°C, and the reaction monitored. Upon the disappearance of the starting material the solution was acidified 20 (aqueous HCl dropwise), and extracted into ethyl acetate (3 x 30ml). The organics were pooled, dried over magnesium sulphate, and the solvent removed under reduced pressure. This yielded a white crystalline solid.

25 EXAMPLE 133-(Phenethyl)oxybenzamide (Compound NU1048)

30 3-Hydroxybenzamide (500mg; 3.6mM) was dissolved in dry acetonitrile (36ml), and to this was added potassium carbonate (0.503g; 3.6mmol) and 2-(bromoethyl)benzene (0.498ml; 3.6mmol). The mixture was heated under reflux for two days. The acetonitrile was removed under reduced pressure to yield a white solid. This was dissolved in 35 water and extracted into dichloromethane (3 x 30ml). The organics were pooled, dried over magnesium sulphate, filtered and the solvent was removed under reduced pressure to yield a white solid. This was recrystallised from boiling ethyl acetate and petrol (0.459g; 12.51%).

MPt : 132-136°C.  $^1\text{H}$ : 200MHz CDCl<sub>3</sub> 3.05 (2H, t, H<sub>8</sub>); 4.2 (2H, t, H<sub>7</sub>); 6.00 (1H, br s, NH); 7.04 (1H, m, H<sub>4</sub>); 7.27 (9H, m, NH; aromatic). IR cm<sup>-1</sup> 3300 (NH<sub>2</sub>); 1666 (CO); 2930 (C=C). M/Z (EI): 241 (M<sup>+</sup>; 18%); 105 (-COHN<sub>2</sub>; 100%).

5

**Elemental Analysis:**

Expected % C<sub>15</sub>H<sub>15</sub>O<sub>2</sub>N C: 74.4, H: 6.20, N: 5.79;  
Found: % C: 74.7, H: 6.2, N: 5.8.

10

Example 14

3-Allyloxybenzamide (Compound NU1031)

15 To 3-hydroxybenzamide (274mg; 2.0mM) under a nitrogen atmosphere was added potassium carbonate (276mg; 21.0mM). This was dissolved in acetonitrile (20ml) containing allyl bromide (169ul; 2.0mM). The mixture was refluxed for 5 hours, and the reaction followed by T.L.C. : 10% MeOH: 90% CH<sub>2</sub>Cl<sub>2</sub>. The excess solvent was then removed under vacuum, the organics extracted into dichloromethane, dried, and the solvent removed, to yield a white solid. This was recrystallised from hot water to produce a white "needle" solid (63%) as the final product.

20 25 Melting point: 116-117°C

$^1\text{H}$  NMR: δ=4.7(m; 2H; H<sub>e</sub>, H<sub>f</sub>); 5.5(m; 2H; H<sub>h</sub>, H<sub>i</sub>); 6.2(m; 1H; H<sub>g</sub>); 7.2(m; 1H; H<sub>d</sub>); 7.6(m; 4H; H<sub>a</sub>, H<sub>b</sub>, H<sub>c</sub>, NH); 8.1(s; 1H; NH).

$^{13}\text{C}$  NMR: δ=68.580, 113.821, 117.978, 129.661, 133.893, 136.025, 158.371, 167.892.

30 Mass spectra: EI: 177 (M<sup>+</sup>); 41 (100%)

EXAMPLE 15

35 3-(Cinnamyl)benzamide (Compound NU1050)

3-hydroxybenzamide (0.5g 3.6mmol) was dissolved in dry acetonitrile (50ml). To this was added potassium carbonate (0.503g; 3.6mmol) and cinnamyl chloride (0.5ml;

3.6mmol). This was left to reflux for 3.5 hours. The acetonitrile was removed under reduced pressure yielding a white solid which was dissolved in water (80ml). The organics were extracted in dichloromethane (3 x 30ml),  
5 dried over magnesium sulphate, filtered and the solvent removed under reduced pressure leaving a white solid. The solid was recrystallised from ethyl acetate and petrol (45%).

10 Mpt : 131-138°C. M<sub>r</sub> 137 (36%); 165 (100%, cinnamyl<sup>+</sup>),  
94 (48%); 77 (24%). IR cm<sup>-1</sup> 3400 & 3200 (NH<sub>2</sub>), 1600 (C=O).  
<sup>1</sup>H: 200MHz d<sub>6</sub>DMSO 4.88 (2H, d, H<sub>7</sub>); 6.65 (1H, m, H<sub>8</sub>); 6.94  
(1H, m, H<sub>4</sub>); 7.24 (1H, m, H<sub>4</sub>); 7.5 (9H, m, aromatic & NH);  
8.1 (1H, s, NH).

15

Elemental Analysis:

Expected C: 75.9%; H: 5.9%; N: 5.5%;

Found C: 76.07%; H: 5.85%; N: 5.56%.

20

EXAMPLE 16

2-Methylbenzoxazole-4-carboxamide (Compound NU1056)

25

1st Stage - Preparing Methyl (3-Hydroxy-2-nitro)benzoate

30 3-hydroxy-2-nitrobenzoic acid (5g; 27.32mM) was dissolved in anhydrous methanol (200ml). Anhydrous hydrogen chloride gas was bubbled through the solution until saturated. The mixture was then refluxed for 20 hours (reaction followed by T.L.C. : 10% methanol/90% dichloromethane). Next, the solvent was removed under vacuum to yield a brown solid. The solid was dissolved in water (100ml) and sodium bicarbonate was added until effervescence stopped. Sodium chloride (15g) was added to the aqueous solution, and the product was extracted into ethyl acetate (3x50ml). The pooled aliquots were dried over magnesium sulphate and the solvent removed under vacuum to yield a malty brown solid.  
35

T.L.C. (as before): r.f.: 0.53

NMR: 200MHz:  $d_6$  DMSO:  $\delta$ =3.9 (s;3H; $OCH_3$ ); 7.1(m;1H; $H_4$ );  
7.35(m;1H; $H_6$ ); 7.8(t;1H; $H_5$ )

Yield 92%

5

2nd Stage - Preparation of Methyl (2-Amino-3-hydroxy)-benzoate

10 Under a nitrogen atmosphere a palladium/carbon catalyst was suspended in anhydrous methanol (150ml). To this suspension was added methyl (3-hydroxy-2-nitro) benzoate from Stage 1 (4g; 20.3mM). The mixture was left under a hydrogen atmosphere for 4½ hours. The catalyst was  
15 removed by filtration through a "celite" pad, and the solvent removed from the filtrate to yield an orange/brown product.

T.L.C.: 40% Ethyl acetate/60% petroleum ether 60/80.

r.f. 0.33

20 NMR 200MHz:  $d_6$ DMSO :  $\delta$ : 3.81(S;3H; $OCH_3$ );  
6.2(broad;2H; $NH_2$ ); 6.5(t;1H; $H_5$ ); 6.9(m;1H; $H_6$ )  
7.3(m;1H; $H_4$ ) 9.8(s;1H;OH)

Yield 83%

25 3rd Stage - Preparation of Methyl 2-methylbenzoxazole-4-carboxylate

To a solution of the methyl (2-amino-3-hydroxy)-benzoate (3g; 18.01mM) in m-xylene (150ml) was added acetyl chloride (1.518ml; 21.6mM). A precipitate was formed; this was left to stir for 30 minutes. On the addition of triethylamine (2.97ml; 21.6mM) the solution became translucent. Pyridinium-p-toluene sulphonic acid (1.2g; 21.6mM) was added, and the mixture refluxed for 34 hours.  
35 The solvent was removed by distillation (vacuum) to yield a brown solid which was column chromatographed (50% ethyl acetate/50% petrol 60/80) to give the desired product as a yellow solid.

T.L.C. As above.

<sup>1</sup>H NMR : 200MHz; CDCl<sub>3</sub>; δ: 2.69(3H;s;2-CH<sub>3</sub>)  
3.99(3H;s;OMe); 7.33(3H;t;H<sub>6</sub>); 7.63(1H;dd;H<sub>7</sub>)  
7.94(1H;dd;H<sub>5</sub>)

5

4th Stage - Preparation of 2-methylbenzoxazole-4-carboxylic acid

Methyl 2-methylbenzoxazole-4-carboxylate (0.1 g,  
10 0.523 mmol) was dissolved in methanol (3 ml), and to this  
was added aqueous sodium hydroxide solution (0.2 M, 3 ml).  
The mixture was stirred at 40°C for 4 hours and acidified  
15 with hydrochloric acid (6 M) until pH = 1.0. The mixture  
was extracted with ethyl acetate (3 x 20 ml), the combined  
organic layers were washed with water (2 x 20 ml), dried  
(MgSO<sub>4</sub>) and the solvent was removed under reduced pressure  
to afford the carboxylic acid (0.068 g, 73%).

20

5th Stage - Preparation of 2-methylbenzoxazole-4-carboxamide

A solution of 2-methylbenzoxazole-4-carboxylic acid  
25 (0.1 g, 0.28 mmol) in anhydrous THF (10 ml) was stirred  
under nitrogen, and thionyl chloride (0.022 ml, 0.31 mmol),  
and DMF (0.1m) were added, whereupon the mixture was  
stirred for a further 5 hours at room temperature. Aqueous  
ammonia (0.5 ml) was added and the mixture was stirred for  
30 a further 30 minutes. The solvent was removed under  
reduced pressure, the residual solid was dissolved in water  
(20 ml), and the solution was extracted with ethyl acetate  
(3 x 20 ml). The organic layers were pooled, washed with  
water (2 x 20 ml) and dried (MgSO<sub>4</sub>). The solvent was  
35 removed under reduced pressure to furnish the carboxamide  
(0.083 g, 84%).

<sup>1</sup>H (200 MHz) CDCl<sub>3</sub> δ = 6.0 (brs, 1H, NH), 7.4 (t, 1H, H<sub>6</sub>),  
7.6 (dd, 1H, H<sub>7</sub>), 8.15 (dd, 1H, H<sub>5</sub>), 8.8 (brs, 1H, NH).

EXAMPLE 16a2-Methylbenzoxazole-4-carboxamide (Compound NU1056)

5 In a modification of the procedure described under Example 16 above, the product of the 3rd stage, methyl 2-methylbenzoxazole-4-carboxylate, was prepared directly from the product of the 1st stage, as described below.

10 Methyl 3-hydroxy-2-nitrobenzoate (0.1g; 0.59mmol) from the 1st stage was dissolved in anhydrous ethanol (20ml) and to this was added ethyl acetimidate hydrochloride (0.067g; 0.59mmol). The reaction mixture was then heated under reflux for 24 hours. The ethanol was removed  
15 under reduced pressure yielding a brown crystalline solid. This was dissolved in ethyl acetate (3 x 20ml) to produce a precipitate of excess ethyl acetimidate hydrochloride. The excess imidate was filtered off and the solution washed with sodium hydroxide solution (0.1N; 3 x 20ml), and water  
20 (3 x 50ml). The solvent was dried over magnesium sulphate and removed under reduced pressure, leaving a orange crystalline solid (0.1265g; 85%).

EXAMPLE 17

25

2-t-butylbenzoxazole-4-carboxamide (Compound NU1040)(a) 1st Stage - Preparation of Methyl 2-t-butyl-  
benzoxazole-4-carboxylate

30

35 2-Amino-3-hydroxybenzoate (0.1g; 0.598mmol) was dissolved in m-xylene and warmed to 70°C. To this was added pivaloyl chloride (0.117ml; 0.958mmol), whereupon a brown precipitate was observed to develop. The mixture was stirred for 30 minutes before the addition of triethylamine (0.099ml; 0.958mmol) and pyridinium-4-toluenesulphonate (0.04g; 0.958mmol). The mixture was then heated under reflux for 26 hours. The m-xylene was removed under reduced pressure to yield a sticky brown solid. The solid

was dissolved in water (50ml) and the organics extracted into ethyl acetate (3 x 30ml), pooled, dried over magnesium sulphate, filtered and the solvent removed under reduced pressure.

5

The title compound was purified via silica column chromatography, with 1:1 ethyl acetate:petrol as the eluant to yield a yellow solid (69%).

10

IR  $\text{cm}^{-1}$ : 3040 (3 x  $\text{CH}_3$ ), 1709 ( $\text{C}=\text{O}$ ). M/Z: 233 (63%;  $\text{M}^+$ ); 218 (50%;  $-\text{CH}_3$ ) 202 (42%;  $-\text{OCH}_3$ ); 186 (100%); 173 (12%);  $-\text{CH}_3$ ); 160 (33%); 146 (62%); 117 (43%).  $^1\text{H}$ :  $d_6$  DMSO:  $\delta$  = 1.17 (9H; s;  $(\text{CCH}_3)_3$ ); 3.99 (3H; s,  $\text{OCH}_3$ ); 7.31 (1H; t;  $\text{H}_6$   $J=8\text{Hz}$ ), 7.66 (1H; dd;  $\text{H}_7$ ;  $J=7.8$ , 1Hz), 7.95 (1H; dd;  $\text{H}_5$   $J=7$ , 1Hz).

15

(b) 2nd Stage - Preparation of 2-t-Butylbenzoxazole-4-carboxamide

20

Methyl 2-t-butyl-4-benzoxazole carboxylate (0.1g; 0.46mmol) was dissolved in methanol (5ml), and to this was added aqueous ammonia (5ml). The mixture was warmed to 40°C and left to stir for 6 hours at ambient temperature. Once the reaction was complete the solvent was removed under reduced pressure and the product was recrystallised from boiling ethyl acetate and petrol (73%).

30

IR  $\text{cm}^{-1}$ : 3395; 3304 (amide NH); 3163 (3 x  $\text{CH}_3$ ), M/Z: 218 (96%;  $\text{M}^+$ ); 202 (43%;  $-\text{NH}_2$ ) 186 (85%;  $-\text{CONH}_2$ ); 175 (77%); 160 (35%) 146 (79%); 133 (23%); 41 (100%).

$^1\text{H}$ :  $\text{CDCl}_3$ :  $\delta$  = 1.44 (9H; s;  $(\text{CCH}_3)_3$ ); 5.95 (1H; br s; NH); 7.34 (1H; t;  $\text{H}_6$   $J=8\text{Hz}$ ), 7.6 (1H; dd;  $\text{H}_7$ ;  $J=7.8$ , 2Hz) 8.07 (1H; dd;  $\text{H}_5$   $J=6.7$ , 2Hz); 8.88 (1H; br s; NH)

35

$^{13}\text{C}$  28.416 (3x $\text{CH}_3$ ), 34.372 ( $\text{CCH}_3$ ), 113.905 (Ar), 123.277 (Ar), 124.338 (Ar), 124.338 (Ar), 125.440 (Ar), 139.371 (O-Ar), 150.804 (N-Ar), 166.457 (Ar), 174.471 ( $\text{C}=\text{O}$  amide).

CHN: Found: C 65.915%; H 6.39%; N 12.48%, Required: C 66.038%; H 6.465%; N 12.835%.

EXAMPLE 182-Phenylbenzoxazole-4-carboxamide (Compound NU1051)5 (a) 1st Stage - Preparation of Ethyl benzimidate Hydrochloride

Benzonitrile (0.514ml; 5mmol) was added to anhydrous ethanol (0.69g). Anhydrous hydrogen chloride gas was bubbled through the solution until saturated. The mixture was left to stir for 20 hours. The white crystalline solid was collected and dried.

mpt: 125-130°C

15 IR : cm<sup>-1</sup>, 2856, 1631. M/Z: 148 (M<sup>+</sup>, 30%), 105 (100%);

<sup>1</sup>H: d<sub>6</sub> DMSO; δ = 1.5 (3H; t, J=7Hz; CH<sub>2</sub>CH<sub>3</sub>), 4.75 (2H; q, J=6.9Hz; CH<sub>2</sub>CH<sub>3</sub>), 8.0 (5H; m; aromatics).

20 Elemental Analysis:

Expected C: 58.25; H: 6.51; N: 7.54;

Found C: 58.13; H: 6.43; N: 7.36.

25 30 (b) 2nd Stage - Preparation of Methyl 2-phenylbenzoxazole-4-carboxylate

To Methyl 2-amino-3-hydroxybenzoate (0.10g; 0.59mmol) was added ethyl benzimidate hydrochloride (0.167g; 0.998mmol) in anhydrous ethanol (20ml). This was refluxed for 20 hours, and the reaction followed by TLC (1:4 Ethyl acetate:petrol).

35 The ethanol was removed under reduced pressure and the solid dissolved in water (20ml), the organics were extracted into ethyl acetate (3 x 20ml), pooled, and washed with sodium hydroxide solution (2 x 10ml; 0.2M), water (2 x 10ml) and then dried over magnesium sulfate. The solvent was removed under reduced pressure to yield a yellow solid.

The title compound was isolated by flash chromatography (eluant as for TLC), yielding an off-white crystalline solid. (85%).

5 IR : cm<sup>-1</sup>, 1714. <sup>1</sup>H: d<sub>6</sub> DMSO: δ = 4.1 (3H; s; OCH<sub>3</sub>), 7.39 (1H; t; J=6Hz; H<sub>6</sub>), 7.5 (3H; m; H<sub>3'</sub>; H<sub>4'</sub>; H<sub>5'</sub>), 7.78 (1H; dd J=7.0, 1.0 Hz; H<sub>7</sub>), 8.0 (1H; dd; J= 6.67 & 1.14Hz; H<sub>5'</sub>), 8.3 (2H; m; H<sub>2'</sub>; H<sub>6'</sub>).

10

(c) 3rd Stage - Preparation of 2-phenylbenzoxazole-4-carboxamide

To a solution of methyl 2-phenylbenzoxazole-4-15 carboxylate (0.02g; 0.0905mmol) in methanol (3ml) was added aqueous ammonia (3ml). This was warmed to 40°C and left to stir, whilst the reaction was monitored by TLC (1:4 EtOAc:petrol). Once all the starting material had reacted a white precipitate was formed. The solvent was removed to 20 yield a white solid, which was recrystallised from ethyl acetate and petrol (70%).

MPt: 199-201°C.

25 <sup>1</sup>H: CDCl<sub>3</sub>: δ = 6.02 (1H; br s; NH); 7.43 (1H; t; H<sub>6</sub>; J=8Hz), 7.57 (3H; m; H<sub>3'</sub>; H<sub>4'</sub>; H<sub>5'</sub>), 7.74 (1H; dd; H<sub>7</sub>; J=1Hz, & 7Hz); 8.23 (1H; dd; H<sub>5</sub>; J=1Hz & 7.3Hz); 8.27 (2H; m; H<sub>2'</sub>; H<sub>6'</sub>), 8.97 (1H; br s; NH).  
13C ; M/Z: (EI); 238 (m<sup>+</sup>; 100%); 222 (-NH<sub>2</sub> 68%);  
30 195 (-CONH<sub>2</sub>; 98%). IR cm<sup>-1</sup> 3383; 3165.

Elemental Analysis:

Expected C:70.58, H:4.23, N:11.76.

Found C:70.41, H:4.24, N:11.77.

EXAMPLE 192-(4-Nitrophenyl)benzoxazole-4-carboxamide (Compound NU1053)

5

(a) 1st stage - Preparation of Methyl 3-hydroxy-2-(N-4-nitrobenzoyl)aminobenzoate

Methyl 2-amino-3-hydroxybenzoate (0.5g; 2.99mmol) was dissolved in m-xylene (40ml) with warming to 60°C. 4-Nitrobenzoyl chloride (0.556ml; 2.99mmol) was added dropwise, and this was left to stir for 4 hours. The solution was cooled to ambient temperature and the m-xylene removed under reduced pressure. The solid was dissolved in water (100ml) and the organics extracted into ethyl acetate (3 x 50ml). The organic fractions were pooled, dried over magnesium sulphate, filtered and the solvent removed under reduced pressure.

The title product was purified via column chromatography (1:4 ethyl acetate:petrol as eluent) to yield an orange solid (49%).

IR : cm<sup>-1</sup>: 3443 (OH), 2953, 1697, 1649, 1404. M/Z: 316 (15% M<sup>+</sup>).

<sup>1</sup>H: CDCl<sub>3</sub>: δ = 3.95 (3H; s; OCH<sub>3</sub>), 7.25 (1H; t; H<sub>5</sub> J=8Hz), 7.31 (1H; dd; H<sub>4</sub>,; J=6,& 2Hz), 7.67 (1H; dd; H<sub>6</sub>,); 8.26 (2H; dd; H<sub>2</sub>'; H<sub>6</sub>' J=2.3Hz); 8.30 (2H; dd; H<sub>3</sub>'); H<sub>5</sub>' J=2.2Hz); 9.81 (1H; s; OH); 12.30 (1H; s; NH).

30

(b) 2nd Stage - Preparation of Methyl 2-(4-nitrophenyl)-benzoxazole-4-carboxylate

35

Methyl 3-hydroxy-2-(N-4-nitrobenzoyl)aminobenzoate (0.1g; 0.34 mmol) was dissolved in m-xylene (20ml), and to this was added triethylamine (0.033ml; 0.45mmol) and pyridinium-4-toluene sulphonate (0.070g; 0.28mmol). This was refluxed for 32 hours. The m-xylene was removed under

reduced pressure and the remaining solid dissolved in water. The organics were extracted into ethyl acetate (3 x 30ml), dried, filtered and the solvent removed under reduced pressure to yield a brick red solid (74%).

5

IR :  $\text{cm}^{-1}$ : 1726; 1522; 1556, M/Z: 298 (84%,  $M^+$ ) 267 (100%, -OCH<sub>3</sub>), 240 (-CO) <sup>1</sup>H: CDCl<sub>3</sub>: δ = 4.01 (3H; s; OCH<sub>3</sub>), 7.44 (1H; t; H<sub>6</sub> J=8.1Hz), 7.76 (1H; dd; H<sub>7</sub>; J=7.2, & 1Hz), 8.04 (1H; dd; H<sub>5</sub>), 8.35 (2H; dd; H<sub>2'</sub>; H<sub>6'</sub> J=2.2Hz); 8.46 10 (2H; d; H<sub>3'</sub>; H<sub>5'</sub> J=2.2Hz).

(c) 3rd Stage - Preparation of 2-(4-nitrophenyl)benzoxazole-4-carboxylic acid

Methyl 2-(4-nitrophenyl)benzoxazole-4-carboxylate 15 (0.1 g, 0.335 mmol) was dissolved in methanol (3 ml), and to this was added aqueous sodium hydroxide solution (0.2 M, 3 ml). The mixture was stirred at 40°C for 4 hours and acidified with hydrochloric acid (6 M) until pH = 1.0. The mixture was extracted with ethyl acetate (3 x 20 ml), the 20 combined organic layers were washed with water (2 x 20 ml), dried ( $\text{MgSO}_4$ ) and the solvent was removed under reduced pressure to afford the carboxylic acid (0.084 g, 89%).

25

(d) 4th Stage - Preparation of 2-(4-nitrophenyl)benzoxazole-4-carboxamide

A solution of 2-(4-nitrophenyl)benzoxazole-4-carboxylic acid (0.084 g, 0.29 mmol) in anhydrous THF (10 ml) was stirred under nitrogen, and thionyl chloride (0.022 ml, 0.31 mmol), and DMF (0.1ml) were added, whereupon the mixture was stirred for a further 5 hours at room temperature. Aqueous ammonia (0.5 ml) was added and the mixture was stirred for a further 30 minutes. The solvent was removed under reduced pressure, the residual solid 30 dissolved in water (20 ml), and the solution was extracted with ethyl acetate (3 x 20 ml). The organic layers were pooled, washed with water (2 x 20 ml) and dried ( $\text{MgSO}_4$ ). The solvent was removed under reduced pressure to furnish 35 the carboxamide (0.07 g, 85%).

EXAMPLE 20

2-(4-Methoxyphenyl)benzoxazole-4-carboxamide (Compound  
NU1054)

5

(a) 1st Stage - Preparation of Methyl 3-hydroxy-2-(N-4-methoxybenzoyl)aminobenzoate

Methyl 2-amino-3-hydroxybenzoate (0.5g; 2.99mmol) was  
10 dissolved in m-xylene (40ml) with warming to 60°C. 4-Methoxybenzoyl chloride (0.509g; 2.99mmol) was added dropwise, and this was left to stir for 3 hours. The solution was cooled to ambient temperature and the m-xylene removed under reduced pressure. The solid was dissolved in  
15 water (100ml) and the organics extracted into ethyl acetate (3 x 50ml). The organic fractions were pooled, dried over magnesium sulphate, filtered and the solvent removed under reduced pressure.

The title product was purified via chromatography (1:4  
20 ethyl acetate:petrol as eluent) and recrystallised from boiling ethyl acetate/petrol to yield a brick red solid (33%).

IR :  $\text{cm}^{-1}$ : 3100 (OH) 2571, 1691, 1643, 1606, M/Z: 301 (23%,  
M<sup>+</sup>), 270 (-OCH<sub>3</sub>), 135 (100%, COPhOCH<sub>3</sub>)  
25 <sup>1</sup>H: CDCl<sub>3</sub>: δ = 3.87 (3H; s; OCH<sub>3</sub>); 3.93 (3H; s; COOCH<sub>3</sub>),  
7.02 (2H; dd; H<sub>5'</sub>; H<sub>3'</sub>); 7.14 (1H; t; H<sub>5</sub> J=8Hz), 7.29  
(1H; dd; H<sub>4</sub>,; J=6.3,& 1.7Hz), 7.63 (1H; dd; H<sub>6</sub>); 8.03  
(2H; dd; H<sub>2'</sub>; H<sub>6'</sub>); 10.38 (1H; s; OH); 11.98 (1H; s;  
NH).

30

(b) 2nd Stage - Preparation of Methyl 2-(4-methoxyphenyl)benzoxazole-4-carboxylate

Methyl 3-hydroxy-2-(N-4-methoxybenzoyl)aminobenzoate  
35 (0.05g; 0.166mmol) was dissolved in m-xylene (20ml), and to this was added triethylamine (0.016ml; 0.215mmol) and pyridinium-4-toluene sulphonate (0.034g; 0.13mmol). This was refluxed for 58 hours. The m-xylene was removed under reduced pressure and the remaining solid dissolved in

water. The organics were extracted into ethyl acetate (3 x 30ml), dried, filtered and the solvent removed under reduced pressure to yield a solid (74%).

5 IR cm<sup>-1</sup>: 1718, 1614, 1502. M/Z: 283 (45%, M<sup>+</sup>) 252 (31%, - OCH<sub>3</sub>), 225, 63 (100%).

<sup>1</sup>H: CDCl<sub>3</sub>: δ = 3.88 (3H; s; COOCH<sub>3</sub>); 6.9 (2H; d; H<sub>3'</sub>; H<sub>5'</sub>); 7.35 (1H; t; H<sub>6</sub>), 7.74 (1H; dd; H<sub>7'</sub>), 7.96 (1H; dd; H<sub>5'</sub>); 8.2 (2H; d; H<sub>2'</sub>; H<sub>6'</sub>).

10

(c) 3rd Stage - Preparation of 2-(4-methoxyphenyl)benzoxazole-4-carboxamide

Methyl 2-(4-methoxyphenyl)benzoxazole-4-carboxylate 15 was dissolved in liquid ammonia (30ml) and sealed in an autoclave. The reaction mixture was left at 55°C, 20bar for >20 hours. Once the reaction was complete the ammonia was removed and the resulting solid recrystallised from boiling ethyl acetate and petrol.

20

EXAMPLE 21

25 3-(5-Bromopentyloxy)benzamide (Compound NU1019)

A mixture of 3-hydroxybenzamide (0.5g, 3.65mmol), 1,5-dibromopentane (1.82g, 1.1ml, 7.3mmol) and potassium carbonate (500mg, 3.65mmol) was refluxed (2h) in acetonitrile (18ml) until the reaction was complete by TLC. The solvent was then removed by rotary evaporation to leave a white sticky solid which was chromatographed (10% methanol in dichloromethane on silica) to give a white solid. This was recrystallised from a mixture of petrol and ethyl acetate to give white flaky crystals (0.744g, 2.6mmol, 70% yield). mp 98-99°C.

<sup>1</sup>H (200MHz, d<sub>6</sub>-DMSO): 8.09 (1H, s, NH), 7.49 (4H, m, H<sub>2</sub>, 5, 6 and NH), 7.1 (1H, d, J 7.99, H<sub>4</sub>), 4.11 (2H, t, J 6.2, CH<sub>2</sub>O), 3.67

(2H,t,J 6.6,2H), 1.90 (4H,m,CH<sub>2</sub>CH<sub>2</sub>CH<sub>2</sub>), 1.62 (2H,m,CH<sub>2</sub>); δ<sub>C</sub>(200 MHz, d<sub>6</sub>-DMSO): 168(CO), 158.85(C3), 129.63(C2), 120(C4), 117.79(C6), 113.53(C5), 67.73(OCH<sub>2</sub>), 35.36(CH<sub>2</sub>Br), 32.30(CH<sub>2</sub>CH<sub>2</sub>O), 28.12 (CH<sub>2</sub>CH<sub>2</sub>Br), 24.63 (OCH<sub>2</sub>CH<sub>2</sub>CH<sub>2</sub>); 5 M/Z (EI) 285(M<sup>+</sup>).

**Elemental Analysis:**

Found: C: 50.71%, H: 5.36%, N: 4.95%. C<sub>15</sub>H<sub>22</sub>NO<sub>2</sub>Br

Requires C: 50.366%, H: 5.636%, N: 4.895%.

10

**EXAMPLE 22**

15 3-(8-N9-Adenine-octyloxy)benzamide (Compound NU1022)

A bomb reaction vessel was charged with a mixture (110mg, 0.27mmols) of 3-[8-N7-(6-chloropurine)octyloxy]-benzamide, 3-(8-N9-(6-chloropurine)octyloxy)benzamide and liquid ammonia and heated under pressure (64°C, 30 bar, 20 16H). The ammonia was evaporated to leave a yellow solid which was chromatographed (silica, 15% methanol/dichloromethane) to give a pale yellow solid which was recrystallised from methanol (56mg, 0.17mmol, 62% yield)

mp 199-199.5°C.  
 25 v cm<sup>-1</sup>; δ<sub>H</sub> (200MHz, d<sub>6</sub>-DMSO) 8.24 (1H,s,purine H8), 8.23 (1H,s,purine H2), 8.06 (1H,s,CONH), 7.55-7.39 (4H,m,CONH and aromatic H2,5,6), 7.30 (2H,s,NH<sub>2</sub>), 7.17-7.13 (1H,m,aromatic H4), 4.23 (2H,t,J 7,OCH<sub>2</sub>), 4.07 (2H,t,J 6.3,CH<sub>2</sub>N), 1.94-1.76 (4H,m,OCH<sub>2</sub>CH<sub>2</sub> and CH<sub>2</sub>CH<sub>2</sub>N), 1.40 (8H,m,alkyl env); δ<sub>C</sub>(200MHz, d<sub>6</sub>-DMSO) 167.998 (CO), 158.847 (C3), 152.711 (pC8), 141.206 (pC2), 136.047 (C1), 129.667 (C5), 119.967 (C6), 117.866 (C4), 113.592 (C2), 67.900 (OCH<sub>2</sub>), 43.221 (NCH<sub>2</sub>), 29.698 (OCH<sub>2</sub>CH<sub>2</sub>), 28.961 (OCH<sub>2</sub>CH<sub>2</sub>CH<sub>2</sub> and NCH<sub>2</sub>CH<sub>2</sub>), 26.313(OCH<sub>2</sub>CH<sub>2</sub>CH<sub>2</sub>CH<sub>2</sub>), 25.762(NCH<sub>2</sub>CH<sub>2</sub>CH<sub>2</sub>CH<sub>2</sub>); m/z 338(M<sup>+</sup>).

**Elemental Analysis:**

Found: C,62.42, H,6.48, N,21.89 C<sub>20</sub>H<sub>26</sub>N<sub>6</sub>O<sub>2</sub>

Requires: C,62.81, H,6.85, N,21.97%.

EXAMPLE 233-[5-(6-Chloropurin-9-yl)pentyloxy]benzamide (Compound NU1023)

5

A solution of 3-(bromopentyloxy)benzamide (500mg, 1.7mmols), 6-chloropurine (270mg, 1.7mmols) and potassium carbonate (240mg, 1.7mmols) in DMF (7.5ml) was stirred for two days at room temperature. The solvent was then removed under vacuum (0.001mmHg) and the white solid (mixture of N7 and N9 isomers) was chromatographed (silica, 10% MeOH/CH<sub>2</sub>Cl<sub>2</sub>) to give a single product (N9 isomer) as a glass. This was triturated with ether (~ 5ml) to give a white solid which was recrystallised from ethyl acetate to give the product (120mg, 0.33mmols, 20% yield)

mp 132-133°C

$\nu$  cm<sup>-1</sup>;  $\delta_H$  (200MHz, d<sub>6</sub>-DMSO) 8.88(1H,s,purine H8), 8.85(1H,s,purine H2) 8.05(1H,s,CONH). 7.56(4H,m,NH and aromatic H2,5,6), 7.15-7.10(1H,m,H4), 4.44(2H,t,J7,OCH<sub>2</sub>), 4.08(2H,t,J6.3,CH<sub>2</sub>N), 2.12-1.98(2H,m,OCH<sub>2</sub>CH<sub>2</sub>), 1.94-1.80(2H,m,NCH<sub>2</sub>CH<sub>2</sub>), 1.59-1.43(2H,m,CH<sub>2</sub>CH<sub>2</sub>CH<sub>2</sub>);  $\delta_C$  (200MHz, d<sub>6</sub>-DMSO) 167.925(CO), 158.774(C3), 152.306(pC4), 151.760(pC8), 149.293(pC2), 136.019(C1), 131.175(pC6), 129.596(C5), 119.961(C6), 117.801(C4), 113.458(C2), 67.591(OCH<sub>2</sub>), 44.033(NCH<sub>2</sub>), 29.071(OCH<sub>2</sub>CH<sub>2</sub>), 28.337(CH<sub>2</sub>CH<sub>2</sub>N), 22.890(CH<sub>2</sub>CH<sub>2</sub>CH<sub>2</sub>); m/z 360 (M<sup>+</sup>).

30 Elemental Analysis:

Found C, 56.73, H, 5.28, N, 19.4. C<sub>17</sub>H<sub>18</sub>N<sub>5</sub>O<sub>2</sub>Cl

Requires C: 56.57%, H: 5.04%, N: 19.46%.

EXAMPLE 243-(5-Adenos-9-ylpentyloxy)benzamide (Compound NU1024)

5 A bomb reaction vessel was charged with a mixture (392.1mg, 1.1mmols) of 3-[5-N7-(6-chloropurine)pentyloxy]-benzamide, 3-[5-N9-(6-chloropurine)pentyloxy]benzamide and liquid ammonia and heated under pressure (66°C, 30 bar, 16H). The ammonia was evaporated away to leave a yellow  
10 solid that was chromatographed (silica, methanol/dichloro-methane) to give a pale yellow solid which was recrystallised (203.5mg, 0.6mmols, 54% yield).

mp 148-149°C

15

v cm<sup>-1</sup>; δ<sub>H</sub> (200 MHz, d<sub>6</sub>-DMSO) 8.25(2H,m,purine H8,2),  
8.06(1H,s,CONH), 7.61-7.38(4H,m,CONH and aromatic H2,5,6),  
7.31(2H,s,NH<sub>2</sub>), 7.15-7.11(1H,m,H4), 4.27(2H,t,J=7Hz, OCH<sub>2</sub>),  
4.07(2H,t,J=6.3Hz, CH<sub>2</sub>N), 2.05-1.78(4H,m,OCH<sub>2</sub>(CH<sub>2</sub>)<sub>3</sub>CH<sub>2</sub>H),  
20 1.56-1.41(2H,m,NCH<sub>2</sub>CH<sub>2</sub>), 1.59-1.43(2H,m,CH<sub>2</sub>CH<sub>2</sub>CH<sub>2</sub>); δ<sub>C</sub>; m/z (M+).

25 EXAMPLE 25

3-[8-N9-(6-Chloropurine)-octyloxy]benzamide  
(Compound NU1027)

30 A solution of 3-(bromoocetylloxy)benzamide (500mg, 1.5mmols), 6-chloropurine (236mg, 1.5 mmols) and potassium carbonate (210mg, 1.5mmols) in DMF (7.5ml) was stirred for two days at room temperature. The solvent was then removed under vacuum (0.001 mmHg) and the white solid remaining (composed of N7 and N9 isomers) was chromatographed (silica, 5% petrol/THF) to give a single product (N9 isomer) in the form of a white solid (74mg) which was recrystallised (2X) from ethanol to give 58.7mg (10% yield)

mp 116-117°C

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dried (0.001mmHg) to yield the quinazolinone title compound (as shown by X-ray crystallographic analysis).

1 H NMR:  $d_6$ DMSO:  $\delta$ : 2.14(s;3H; $CH_3$ ); 6.87(dd;1H; $H_7$ );  
5 6.94(t;1H; $H_6$ ); 7.29(dd;1H; $H_5$ )

Yield 52%

10 EXAMPLE 27

8-Hydroxy-2-(4-nitrophenyl)quinazolin-4-one  
(Compound NU1057)

15 Methyl 2-(4-nitrophenyl)benzoxazole-4-carboxylate (0.20g) obtained as described in Example 19 (2nd stage) was dissolved in liquid ammonia (30ml) and sealed in an autoclave. The reaction mixture was left at 55°C, 20bar for 20 hours. Under these conditions, the expected 20 2-(4-nitrophenyl)benzoxazole derivative apparently rearranged to give the corresponding quinazolinone derivative. Once the reaction was complete the ammonia was removed and the resulting solid recrystallised from boiling ethyl acetate and petrol (84%).

25

IR  $cm^{-1}$ : M/Z: 283 (38%,  $M^+$ ) 267 (100%,  $-NH_2$ ), 240 (-CO)  
 $^1H$ :  $CDCl_3$ :  $\delta$  = 7.35 (1H; dd;  $H_5$ ), 7.5 (1H; t;  $H_6$ ), 7.7 (1H; dd;  $H_7$ ); 8.45 (2H; d;  $H_2$ ;  $H_6'$ ); 8.79 (2H; d;  $H_3$ ;  $H_5'$ ), 9.98 (1H; br s; NH), 12.8 (1H; br s; NH).

30

35 Similarly, in attempting to prepare benzoxazole-4-carboxamide, the product underwent a molecular rearrangement yielding 8-hydroxy-quinazolin-4-[3H]one (Compound NU1026) which had quite strong PARP inhibitory activity.

Further examples now follow of the preparation of more quinazolinone compounds of particular interest.

EXAMPLE 288-Methoxy-2-methylquinazolin-4-[3H]-one (Compound NU1063)5 Method A(a) 1st Stage - Preparation of 3-Methoxy-2-nitrobenzamide

10 A solution of 3-methoxy-2-nitrobenzoic acid (0.1g, 5.1 mmol), thionyl chloride (0.55ml, 7.6 mmol) and dimethylformamide (0.2ml), in THF (10ml) was stirred for 12 hours at 25°C under nitrogen. Aqueous ammonia (6ml) was cautiously added and the mixture was stirred for a further 15 minutes, the solvent was removed under reduced pressure and the remaining solid was washed with ice-cold water and collected (0.74g, 75%) m.p. 219-222°C

15  $\delta_H$  (200MHz,  $d_6$ -DMSO) 4.01 (s, 3H,  $OCH_3$ ); 7.41-7.46 (dd, 1H, Ar-4H); 7.55-7.60 (dd, 1H, Ar-6H); 7.69 7.77 (m, 1H, Ar-5H); 7.84 (br s, 1H, -NH); 8.31 (br s, 1H, -NH); m/z 196 (34.3%, M<sup>+</sup>)

20  $v_{max}/cm^{-1}$  3350 (br), 3180 (br), 3000, 2970, 2920, 2820, 1675.

Elemental analysis: found C 49.03, H 3.93, N 13.97,  
25  $C_8H_8N_2O_4$  requires C 48.98, H 4.11, N 14.28%.

(b) 2nd Stage - Preparation of 2-Amino-3-methoxybenzamide

30 3-Methoxy-2-nitrobenzamide (0.5g, 2.5 mmol) was dissolved in dry methanol (40ml) and hydrogenated using palladium-carbon catalyst (80mg). The catalyst was removed by filtration through Celite, and the residual product (0.35g) was collected and dried (83%) m.p. 145-147°C

35  $\delta_H$  (200MHz,  $d_6$ -DMSO) 3.88 (s, 3H,  $-OCH_3$ ); 6.40 (br s, 2H, -NH<sub>2</sub>); 6.54-6.62 (t, 1H, Ar-5H); 6.96-6.99 (dd, 1H, Ar-4H); 7.23 (br s, 1H, -NH); 7.29-7.33 (dd, 1H, Ar-6H); 7.85 (br s, 1H, -NH); m/z 166 (43.8%, M<sup>+</sup>)  $v_{max}/cm^{-1}$  3480, 3370, 3330, 3150, 2970, 2850, 1680, 1620.

Elemental analysis: found C 57.54, H 5.99, N 16.61,  
C<sub>8</sub>H<sub>10</sub>N<sub>2</sub>O<sub>2</sub> requires C 57.82, H 6.07, N 16.86%.

(c) 3rd Stage - Preparation of 2-N-Acetyl amino-3-methoxybenzamide

To a solution of 2-amino-3-methoxybenzamide (0.5g, 3.5 mmol) in dry THF (15ml), containing pyridine (0.3ml; 3.9 mmol), was added acetyl chloride (0.2ml, 3.3 mmol) in THF (2ml) dropwise, and the reaction mixture was stirred overnight under nitrogen. The solvent was removed under vacuum and the remaining white slurry washed with aqueous sodium bicarbonate solution, filtered and washed with water. The white product (0.19g, 31 %) was collected and dried.

m.p. 243-246°C

<sup>15</sup> δ<sub>H</sub> (200MHz, d<sub>6</sub>-DMSO) 2.05 (s, 3H, -CH<sub>3</sub>); 3.88 (s, 3H, -OCH<sub>3</sub>); 7.14-7.18 (dd, 1H, Ar-4H); 7.21-7.25 (dd, 1H, Ar-6H); 7.33-7.41 (m, 2H, -NH and Ar-5H; 7.53 (br s, 1H, -NH); 9.27 (br s, 1H, -NH); m/z 208 (16.6, M<sup>+</sup>)  
ν<sub>max</sub>/cm<sup>-1</sup> 3420, 3240 (br), 3160, 3020, 2980, 2870, 1660.  
<sup>20</sup> Elemental analysis: found C 56.98, H 5.38, N 12.78, C<sub>10</sub>H<sub>12</sub>N<sub>2</sub>O<sub>3</sub> requires C 57.68, H 5.81, N 13.46%.

(d) Final Stage - Preparation of 8-Methoxy-2-methylquinazolin-4-[3H]-one

2-N-Acetyl amino-3-methoxybenzamide (0.07g, 0.34 mmol) from 3rd stage was dissolved in aqueous sodium hydroxide solution (2% w/v, 2ml) and the solution was stirred for 12 hours at 25°C. The reaction mixture was neutralised with dilute aqueous hydrochloric acid and the resulting white precipitate that was deposited was collected by filtration and washed thoroughly with water. The title compound was recrystallised from ethyl acetate (0.043g, 67%)  
<sup>35</sup> m.p. 202-204°C (sublimes).

EXAMPLE 28a8-Methoxy-2-methylquinazolin-4-[3H]-one (Compound NU1063)5 Alternative Method B

To a mixture of 2-amino-3-methoxybenzamide (1.0g, 6 mmol) from the 2nd stage of Example 28 and pyridine (0.6ml, 7.8 mmol) in dry THF (25ml), was added a solution of acetyl chloride (0.9ml, 13 mmol) in THF (2ml) dropwise, and the mixture was stirred overnight under nitrogen. The solvent was removed under vacuum and the remaining white slurry was resuspended in 2% aqueous sodium hydroxide solution and neutralised with aqueous hydrochloric acid, whereupon a white precipitate formed. The product was collected by filtration and recrystallised from methanol-water (0.915g, 80%) m.p. 202-204°C (sublimes)

20  $\delta_H$  (200MHz,  $d_6$ -DMSO) 2.43 (s, 3H, - $\text{CH}_3$ ); 3.97 (s, 3H, - $\text{OCH}_3$ ); 7.37-7.50 (m, 2H, Ar-6/7H); 7.68-7.73 (dd, 1H, Ar-5H);  $\delta_C$  ( $d_6$ -DMSO); 21.83 (- $\text{CH}_3$ ); 56.05 (- $\text{OCH}_3$ ); 114.96, 116.99 (Ar-67C); 121.95 (C- $\text{CH}_3$ ); 126.5 (Ar-5C); 140.0 (Ar-8AC); 153.26 (Ar-8C); 154.33 (Ar-4aC); 162.04 (C=O);  $m/z$  190 (96.6%,  $M^+$ )  $v_{\max}/\text{cm}^{-1}$  3171, 3034, 2903, 1676, 1620, 25 1574, 1483.

Elemental analysis: found C 62.14, 62.36, H 5.18, 5.29, N 14.23, 14.36;  $C_{10}\text{H}_{10}\text{N}_2\text{O}_2$  requires C 63.15, H 5.30, N 14.73%.

30

EXAMPLE 298-Hydroxy-2-methylquinazolin-4-[3H]-one (Compound NU1025)

35 (Alternative method to Example 26)

A solution of 8-methoxy-2-methylquinazolin-4-[3H]-one (0.7g, 3.7 mmol) from Example 28 in  $\text{BBr}_3$  (1.0 M in  $\text{CH}_2\text{Cl}_2$ ) 8.4ml, 8.4 mmol) was heated under reflux for 24 hours under

nitrogen. Solvents were removed by distillation under vacuum and the remaining residue was hydrolysed with sodium hydroxide solution (10% w/v). Acidification with aqueous hydrochloric acid afforded a white precipitate, which was 5 removed. The filtrate was extracted with ethyl acetate (3 x 30ml), dried ( $MgSO_4$ ) and the solvent was removed under vacuum. Recrystallisation from propan-2-ol-water afforded the target compound (65%) m.p. 253-258°C

10  $\delta_H$  (200MHz,  $d_6$ -DMSO) 2.48 (s, 3H, -CH<sub>3</sub>); 7.22-7.41 (m, 2H, Ar=6/7H); 7.57-7.62 (dd, 1H, Ar-5H); 9.57 (s, 1H, -OH); 12.26 (s, 1H, -NH);  $\delta_C$  ( $d_6$ -DMSO); 21.72 (-CH<sub>3</sub>); 115.78, 118.42 (Ar-6/7C); 121.76 (C-CH<sub>3</sub>); 126.54 (Ar-5C); 138.27 (Ar-8aC) 152.58 (Ar-8C); 152.87 (Ar-4aC); 162.05 (C=O); m/z 15 176 (100%, M<sup>+</sup>);  $\nu_{max}/cm^{-1}$  3320, 3175, 3030, 2900, 2800, 1670.

Elemental analysis: found C 61.39, H 4.54, N 15.88, C<sub>9</sub>H<sub>8</sub>N<sub>2</sub>O<sub>2</sub> requires C 61.36, H 4.58, N15.94%.

20

EXAMPLE 308-Methoxy-2-phenylquinazolin-4-[3H]-one (Compound NU1065)25 Method A(a) 1st Stage - Preparation of 2-N-Benzoylamino-3-methoxybenzamide

30

To a stirred solution of 2-amino-3-methoxybenzamide (0.5g, 3 mmol) from the 2nd stage of Example 28 in dry THF (15ml), containing pyridine (0.3ml, 3.0 mmol), was added benzoyl chloride (0.4ml, 3.3 mmol) in THF (2ml) dropwise.

35 The reaction mixture was stirred under nitrogen at 25°C. The solvent was removed under vacuum to afford a white slurry which was washed with sodium bicarbonate solution, filtered and washed with water. Recrystallisation from methanol-water afforded the title compound (0.2g, 41%) m.p. 176-180°C;

$\delta_H$  (200MHz,  $d_6$ -DMSO) 3.88 (s, 3H, -OCH<sub>3</sub>); 7.24-7.32 (m, 2H, Ar-4/6H); 7.41-7.49 (m, 2H, -NH, Ar-5H); 7.59-7.73 (m, 4H, -NH, Ph-3'/4'H); 8.04-8.08 (dd, 2H, Ph-2'H); 9.85 (s, 1H, -NH); m/z 270 (74.6%, M<sup>+</sup>).

5

(b) 2nd Stage - Preparation of 8-Methoxy-2-phenyl  
quinazolin-4-[3H]-one

10 2-N-Benzoylamino-3-methoxybenzamide (0.2g, 0.74 mmol) was dissolved in aqueous sodium hydroxide solution (2% w/v, 2ml) and the solution was stirred at room temperature for 12 hours. The reaction mixture was neutralised with hydrochloric acid, and the resulting white precipitate that  
15 formed was collected by filtration and recrystallised from methanol/water (0.12g, 65%) m.p. 252-256°C.

20 EXAMPLE 30a

8-Methoxy-2-phenylquinazolin-4-[3H]-one (Compound NU1065)

25 Alternative Method B

To a solution of 2-amino-3-methoxybenzamide (1.0g, 6 mmol) (from the 2nd stage of Example 28) and pyridine (0.6ml, 7.8 mmol) in dry THF (25ml), was added a solution 30 of benzoyl chloride (0.8ml, 6.6 mmol) in THF (2ml) dropwise, and the mixture was stirred overnight under nitrogen. The solvent was removed under vacuum and the remaining white slurry was resuspended in 2% aqueous sodium hydroxide solution and neutralised with aqueous hydrochloric acid, whereupon a white precipitate formed.  
35 The product was collected by filtration and recrystallised from methanol-water (1.1g, 75%) m.p. 252-256°C;

$\delta_H$  (200MHz,  $d_6$ -DMSO) 4.06 (s, 3H, -OCH<sub>3</sub>); 7.47-7.61 (m, 2H, Ar-4/6H); 7.63-7.69 (m, 3H, Ph-3'/4'H); 7.80-7.85 (dd, 1H,

Ar-5H); 8.27-8.32 (m, 2H, Ph-2'H); 12.70 (s, 1H, -NH); m/z 252 (100%, M<sup>+</sup>);  $\nu_{\text{max}}/\text{cm}^{-1}$  3330, 3190, 3170, 3120, 3070, 2950, 2890, 2830, 1660.

5 Elemental analysis: found C 71.38, H 4.39, N 11.17,  
 $\text{C}_{15}\text{H}_{12}\text{N}_2\text{O}_2$  requires C 71.42, H 4.79, N 11.10%.

EXAMPLE 31

10

8-Hydroxy-2-phenylquinazolin-4-[3H]-one (Compound NU1068)

A solution of 8-methoxy-2-phenylquinazolin-4-[3H]-one (0.5g, 2 mmol) from Example 30 or 30a in  $\text{BBr}_3$  (1.0 M in  $\text{CH}_2\text{Cl}_2$ ) (6ml, 6 mmol) was heated under reflux for 24 hours under nitrogen. Solvents were removed by distillation under vacuum and the remaining residue was hydrolysed with sodium hydroxide solution (10% w/v). Acidification with aqueous hydrochloric acid afforded a white precipitate, which was removed. The filtrate was extracted with ethyl acetate (3 x 30ml), dried ( $\text{MgSO}_4$ ) and the solvent was removed under vacuum. Recrystallisation from propan-2-ol afforded the target compound (0.187mg, 67%)  
m.p. 280-284°C;

25

$\delta_{\text{H}}$  (200MHz),  $d_6$ -DMSO) 7.73-7.50 (m, 2H, Ar-6/7H); 7.66-7.72 (m, 4H, Ar-5H, Ph-3'/4'H); 8.51-8.54 (dd, 2H, Ph-2H); 9.75 (bs, 1H, -OH); 12.60 (bs, 1H, -NH);  $\delta_{\text{C}}$  ( $d_6$ -DMSO); 116.01, 118.68 (Ar-6/7C); 122.03 (C-Ph); 127.43-128.76 (Ph-1'/2'/3'/4'C); 137.98 (Ar 8aC); 150.72 (Ar-8C); 153.31 (Ar-4aC); 162.62 (C=O); m/z 238 (100%, M<sup>+</sup>);  $\nu_{\text{max}}/\text{cm}^{-1}$  (approx. values) 3380 (br), 3180, 3120, 3050, 2940, 1640.

Elemental analysis: found C 69.54, H 4.05, N 11.46,  
 $\text{C}_{14}\text{H}_{10}\text{N}_2\text{O}_2$  requires C 70.58, H 4.23, N 11.76%.

EXAMPLE 322,8-Dimethylquinazolin-4-[3H]-one (Compound NU1069)

5 To a solution of 2-amino-3-methylbenzamide (0.5g, 3.3 mmol) (prepared by conventional methods) and pyridine (0.35ml, 4.3 mmol) in dry THF (15ml), was added a solution of acetyl chloride (0.36ml, 5.0 mmol) in THF (2ml) dropwise, and the mixture was stirred overnight under 10 nitrogen. The solvent was removed under vacuum and the remaining white slurry was resuspended in 2% aqueous sodium hydroxide solution and neutralised with aqueous hydrochloric acid, whereupon a white precipitate formed. The solid was collected and recrystallised from methanol-15 water to furnish the required quinazolinone (0.47g, 81%) m.p. 217-220°C;

$\delta_H$  (200MHz,  $d_6$ -DMSO) 2.44 (s, 3H, -CH<sub>3</sub>); 2.57 (s, 3H, -CH<sub>3</sub>); 7.36-7.44 (t, 1H, Ar-6H); 7.68-7.72 (dd, 1H, Ar-7H); 20 7.97-8.01 (dd, 1H, Ar-5H); 12.25 (br s, 1H, -NH); m/z 174 (100%, M<sup>+</sup>);  $\nu_{max}/cm^{-1}$  3325, 3180, 3040, 2990, 2910, 2880, 2795, 1680, 1620.

Elemental analysis: found C 68.76, H 5.57, N 15.90,  
25 C<sub>10</sub>H<sub>10</sub>N<sub>2</sub>O requires C 68.94, H 5.76, N 16.08%.

ASSAY FOR PARP INHIBITORY ACTIVITY

Compounds of the present invention, particularly those detailed in the preceding Examples, have been tested  
5 in vitro for activity as PARP inhibitors using the following methods and materials.

In principle, the PARP assay used relies upon activating endogenous PARP (as hereinafter described) in  
10 cells containing exogenous [<sup>32</sup>P]-NAD<sup>+</sup> introduced therein by suspending the cells in a solution of [<sup>32</sup>P]-NAD<sup>+</sup> to which they have been rendered permeable in an initial pre-treatment step. The poly(ADP-ribose) which is then synthesised by the enzyme can be precipitated by trichloracetic acid (TCA) and the amount of radio-labelled <sup>32</sup>P incorporated therein measured, e.g. using a scintillation counter, to give a measure of the activity of the PARP under the particular conditions of the experiment. By  
15 repeating the experiment following the same procedure, and under the same conditions, in the presence of each compound to be tested the reduction in enzyme activity, representative of the inhibitory effect of the test compound, can then be ascertained from the reduction, if any, of the amount of [<sup>32</sup>P] measured in the TCA precipitated poly(ADP-ribose).  
20  
25

The results of this assay may be expressed in terms of percentage inhibition or reduction in activity for one or more different concentrations of each compound tested,  
30 or it may be expressed in terms of that concentration of the tested compound which reduces the enzyme activity by 50%, i.e. the IC<sub>50</sub> value. Thus, with a range of different compounds a set of comparative values for inhibitory activity can be obtained.  
35

In practice, L1210 murine leukaemia cells were used as the source of the PARP enzyme after being rendered permeable to exogenous [<sup>32</sup>P]NAD by exposure to hypotonic buffer and cold shock. In the preferred technique

developed, which has been found to give exact and reproducible results, a defined amount of a small synthetic oligonucleotide, in particular a single strand oligonucleotide having the palindromic sequence CGGAATTCCG, is introduced into the cell suspension for activating the PARP enzyme. This oligonucleotide sequence snaps back on itself to form a double-stranded molecule with a single blunt end and provides an effective substrate for activation of PARP. Its behaviour as a potent activator of the enzyme was confirmed in the tests carried out.

The experimental protocol adopted, in which a synthetic oligonucleotide as mentioned above is introduced as a specific activator of PARP, discriminates between PARP and other mono-ADP-ribosyltransferases in the cells. Thus, introduction of such synthetic oligonucleotides causes a 5 to 6 fold stimulation in the radioactive label incorporated and this is attributable solely to PARP activity.

Further details of the assay are given below.

#### Materials

The materials used included the following:

##### DTT (Dithiothreitol)

A 100mM (15.4mg/ml) solution (for use as an antioxidant) was made up, divided into 500µl aliquots and stored at -20°C.

##### Hypotonic buffer:

9mM Hepes	(214mg/100ml)
4.5% Dextran	(4.5g/100ml)
4.5mM MgCl <sub>2</sub>	(92mg/100ml)

The above ingredients were dissolved in about 80ml distilled water, pH was adjusted to 7.8 (NaOH/HCl), the solution was then made up to 100ml with distilled water, and stored in a refrigerator. DTT was added to 5mM just before use (50µl/ml).

Isotonic buffer:

5	40mM Hepes	(1.9g/200ml)
	130mM KCl	(1.94g/200ml)
	4% Dextran	(8g/200ml)
	2mM EGTA	(152mg/200ml)
	2.3mM MgCl <sub>2</sub>	(94mg/200ml)
	225mM Sucrose	(15.39g/200ml)

10           The above ingredients were dissolved in about 150ml distilled water, pH was adjusted to 7.8 (NaOH/HCl), the solution was then made up to 200ml with distilled water and stored in a refrigerator. DTT was added to 2.5mM just before use (25μl/ml).

15           NAD

20           NAD was stored as a solid in pre-weighed aliquots at -20°C. From these, solutions of a concentration of approximately 6mM (4-4.5mg/ml) were freshly made up shortly before performing an assay, and the molarity was checked by measuring the optical density (O.D.) at 260nm. The stock solution was then diluted with water to give a concentration of 600μM and a small amount of <sup>32</sup>P labelled NAD was added (e.g. 2-5μl/ml).

25

Oligonucleotide

30           The oligonucleotide having the palindromic sequence CGGAATTCCG, synthesised by conventional means, was vacuum dried and stored as pellets in a freezer. Before use, it was made up to 200μg/ml in 10mM Tris/HCl, pH 7.8, with each pellet being dissolved completely in 50ml of buffer. The solution was then heated to 60°C in a water bath for 15 minutes, and allowed to cool slowly to ensure correct reannealing. After adding 9.5ml of buffer, the concentration was checked by measuring the optical density of a diluted sample at 260nm. The main solution was then diluted to a concentration of 200μg/ml and stored in 500μl aliquots in a freezer, ready for use.

35

TCA

Solutions of TCA (Trichloroacetic acid) were prepared at two concentrations. 10% TCA + 10% sodium pyrophosphate, and 1% TCA + 1% sodium pyrophosphate.

5

Cells

The L1210 cells used as the source of the PARP enzyme were maintained as a suspension culture in RPMI medium + 10% foetal bovine serum + glutamine and antibiotics (penicillin and streptomycin). HEPES and sodium bicarbonate were also added, and the cells were seeded in 100ml of medium such that there would be a concentration of approximately  $8 \times 10^5$ /ml at the time of carrying out an assay.

15

Method

The compounds being tested were generally made up as a concentrated solution in DMSO (Dimethyl sulphoxide). The solubility of the compound was then checked by adding a quantity of the DMSO solution to a quantity of the isotonic buffer, in the required final proportions that were to be used in carrying out the assay, and after an interval the solution was examined under a microscope for any signs of crystals forming.

25

A desired quantity of the cells, ascertained by counting with a haemocytometer, was then centrifuged (1500rpm in a "Europa" model 24M centrifuge for 5 minutes), the supernatant removed, and the pellets obtained were resuspended in 20ml Dul A at 4°C before centrifuging again at 1500rpm and 4°C. After again removing the supernatant, the cells were resuspended at a concentration of  $3 \times 10^7$  cells/ml in ice cold hypotonic buffer and left for 30 minutes on ice. Nine volumes were then added of ice cold isotonic buffer, and the cells, now rendered permeable to exogenous NAD<sup>+</sup>, were then used within the next hour for carrying out an assay. The permeabilisation of the cells may be checked at this stage by adding duplicate aliquots of cells to an equal volume of trypan blue, leaving for 5

30

35

minutes and then counting on a haemocytometer.

The assay was then carried out using for convenience plastic 15ml conical bottomed assay tubes set up in a  
5 shaking water bath at 26°C which is the optimum temperature  
for this enzyme. In a typical assay using the  
oligonucleotide solution at a concentration of 5 $\mu$ g/ml and  
the test compound/DMSO solution at a concentration of 2%,  
and carrying out the assay in quadruplicate, there would  
10 then be placed in each assay tube 5 $\mu$ l of the  
oligonucleotide solution, 50 $\mu$ l of the 600 $\mu$ M NAD + [ $^{32}$ P]-NAD  
solution, 8 $\mu$ l of the test compound/DMSO solution, and 37 $\mu$ l  
of water. Prior to the start of the experiment this  
"cocktail" would be pre-warmed for 7 minutes at 26°C, as  
15 would be also the cell suspension. The reaction would then  
be started by adding 300 $\mu$ l of the cell suspension. The  
reaction would be stopped by adding 2ml of the 10% TCA +  
10% sodium pyrophosphate solution.

20 In addition to the above, six assay tubes would  
usually be set up as blanks, these containing the same  
ingredients as above but, before adding the cell  
suspension, TCA solution is added to prevent any reaction  
from taking place. This enables corrections to be applied  
25 for any non-specific binding of the labelled material to  
the filter used (see below).

After adding the cell suspension at timed intervals  
30 to each of the assay tubes, the 10% TCA + 10% sodium  
pyrophosphate at 4°C was added to each assay tube exactly 5  
minutes after addition of the cell suspension to that tube.  
Then, after leaving the tubes on ice for a minimum time of  
one hour, the contents of each individual tube were  
35 filtered through an individual filter funnel of a suction  
filter apparatus using GF/C filter elements (rough side up)  
wetted with 10% TCA. After filtering the contents of each  
tube and rinsing the filters several times with 1% TCA + 1%  
sodium pyrophosphate solution, the filters were carefully  
removed and dried before being placed in individual

scintillation vials. Four additional scintillation vials were also set up as reference standards containing 10 $\mu$ l of the 600 $\mu$ M NAD + [ $^{32}$ P]-NAD solution, 10ml scintillant then being added to each vial. Counting was carried out for 2 5 minutes on a  $\beta$  counter to obtain measures of the  $^{32}$ P present, and thus the amount of the poly(ADP-ribose) and activity of the PARP enzyme.

10 RESULTS OF IN VITRO PARP INHIBITION STUDIES

Apart from applying the PARP enzyme assay in accordance with the standard procedure outlined above to a range of compounds which have been made in accordance with 15 the present invention, for comparison purposes it was also applied to certain benzamide compounds, in particular 3-hydroxybenzamide, 3-methoxybenzamide and 3-aminobenzamide, that are already known to exhibit certain PARP inhibitory activity. A full tabulated list of the compounds which 20 have been made and/or studied is hereinafter presented in TABLE III, together with the PARP inhibition assay results obtained in different experiments for different concentrations of the compounds when tested using the assay hereinabove described.

25

In reviewing this list, the known PARP inhibitors 3-aminobenzamide, 3-methoxybenzamide and 3-hydroxybenzamide may be regarded as reference compounds. Although there is 30 considerable variation in activity, and in some cases at least the higher concentrations for aqueous solutions of the test compounds could not be achieved because of low solubility, in general the compounds of the present invention which were tested showed a useful degree of 35 activity. Of especial interest were the benzoxazole analogues, particularly those having the reference numbers NU1056, NU1040, NU1051 and NU1054, which showed relatively high inhibitory activity even at low concentrations. Also of especial interest as potent PARP inhibitors are the quinazolinone derivatives which have been mentioned,

particularly compounds NU1025, NU1057, NU1063, NU1068 and NU1069.

In contrast to the results obtained for the compounds 5 of the present invention, which have in many cases showed PARP enzyme inhibitory properties that are well above average and at least comparable with, if not considerably better than, those of other known benzamide PARP inhibitors, various analogous nicotinamide compounds 10 studied showed no, or very poor, inhibitory activity when tested in the same manner at similar concentrations.

15 FURTHER BIOLOGICAL ACTIVITY STUDIES

Again using cultures of the murine leukaemia L1210 cell line, growth inhibition experiments were carried out to assess the cytostatic effects of the compounds and 20 clonogenic survival assays were performed to assess cytotoxicity, especially in relation to use of the compounds in conjunction with DNA damaging cytotoxic agents such as cytotoxic antitumour drugs or high energy radiation. DNA damage and the effect of the PARP 25 inhibitors on the process of DNA strand break formation and repair has also been assessed by carrying out DNA strand break assays and monitoring by alkaline elution in accordance with published techniques.

30 By way of example some further details are given below of studies carried out using the quinazolinine compounds identified by the reference numbers NU1025 and NU1057 (derivable by molecular rearrangement of corresponding benzoxazole compounds) as representative examples of the PARP inhibiting compounds of the present 35 invention, and also using for comparison the known PARP inhibitors 3-aminobenzamide (3AB) and benzamide (BZ) itself. Results of experiments using the alkylating agent temozolomide (TM) are also reported, taking this as a illustrative example of a cytotoxic DNA damaging antitumour

drug, and in some of the studies carried out gamma ray irradiation was used to damage the cells.

In the growth inhibition assays, typically the L1210  
5 cells would be seeded at  $1 \times 10^4$ /ml in triplicate in 24 well multidishes, and 24 hours later the compounds or drugs being tested would be added in selected combinations and concentrations. At this time one set of replicates would be counted using a Coulter counter ( $N_0$ ), and 48 hours later  
10 the remaining samples would be counted ( $N_1$ ). The percentage (%) growth inhibition of drug-treated samples could then be estimated. In drug combination experiments, where evidence of synergistic effects on cell growth or clonogenicity was being sought, a single, fixed concentration  
15 of a cytotoxic drug sample, e.g. temozolomide, would be taken as the control value.

As an illustration of the results obtained, there is  
20 shown in TABLE I at the end of this description the  $IC_{50}$  values of the above-mentioned PARP inhibitors when used alone and in conjunction with a fixed concentration (100 $\mu$ M) of temozolomide, as estimated from the growth inhibition experiments. Although not shown, it may be noted that exposure of the cells to TM alone caused inhibition of cell  
25 growth with an  $IC_{50}$  value of  $361 \pm 25 \mu$ M. Also, it was established that co-exposure of the cells to 100 $\mu$ M TM with increasing concentrations of the PARP inhibitors caused a synergistic increase in growth inhibition throughout a range of concentrations  
30

It will be seen from Table I that 10-20 fold higher concentrations of the compound NU1025 used alone were required to inhibit cell growth than were required when the  
35 compound was used in conjunction with 100 $\mu$ M TM. For example, the  $IC_{50}$  of NU1025 alone was 0.41mM, and this was reduced to 0.04mM in the presence of TM. In comparison, only 2-3 fold differences were obtained with 3AB and BZ, where there was considerable overlap between the growth inhibitory effects of the compounds per se and their

effects in conjunction with TM. An identical rank order was obtained when comparing the effectiveness of the compounds as PARP inhibitors and their ability to inhibit cell growth which at least suggests that PARP function is essential for cell growth.

In the clonogenic survival assays, typically the L1210 cells were exposed to varying concentrations of TM ± a fixed concentration of PARP inhibitor for a fixed time of 10 16 hours, prior to counting and seeding for colony formation in 0.12-0.15% agarose in drug-free medium. After 7-10 days colonies were stained with 0.5mg/ml MTT and counted by eye on a gridded light box. Survival curves were plotted and typical  $DEF_{10}$  values obtained are 15 hereinafter given in Table II ( $DEF_{10}$  being defined as the ratio of the concentration of TM that reduces survival to 10% divided by the concentration of TM that reduces survival to 10% in the presence of a fixed concentration of PARP inhibitor). Each  $DEF_{10}$  value in Table II represents 20 the average ratio ± S.E. (standard error) derived from the averaged 10% survival for TM alone ( $675 \pm 31\mu M$  from 22 independent survival curves) divided by individual 10% survival values from at least 3 independent survival curves 25 performed in the presence of a fixed concentration of inhibitor.

A reasonable correlation was found between growth inhibitory effects and cytotoxic effects for TM alone with 30 an  $IC_{50}$  value of  $361\mu M \pm 25\mu M$  and a  $LD_{50}$  value of  $251 \pm 13\mu M$  respectively, despite differing exposure times (48 hours for growth inhibition and 16 hours for cytotoxicity). TM has a half life in culture of about 40 minutes, and therefore will exert its full effects well before the 35 minimum duration of exposure of either experiment. All compounds potentiated TM cytotoxicity, but NU1025 produced about the same  $DEF_{10}$  values at very much lower concentrations than 3AB and BZ respectively. For example,  $50\mu M$  NU1025 and  $5mM$  3AB gave equivalent  $DEF_{10}$  values of about 4. For NU1025 maximal potentiation of cytotoxicity

was obtained by a concentration in the range of 50-100 $\mu$ M, and was significant at doses as low as 10 $\mu$ M.

In other clonogenic survival assays gamma ray  
5 irradiation was used to damage the cells. Typically, L1210  
cells (3ml, 4 x 10<sup>3</sup>/ml in plastic bijoux bottles) were  
irradiated at 4°C with varying doses of gamma rays in the  
presence or absence of 10mM 3AB or 200 $\mu$ M NU1057 and a final  
concentration of 2% DMSO. The cells were then incubated at  
10 37°C for 2 hours in the continued presence or absence of  
PARP inhibitor prior to seeding for colony formation. A  
significant potentiation of gamma ray cytotoxicity by  
NU1057 was observed, with a DEF<sub>10</sub> of 1.1.

15

Repair of potentially lethal damage (PLD) occurs when  
cells are held in stationary-phase following initiation of  
PLD prior to allowing cell division to take place. In  
further typical experiments, L1210 cells were allowed to  
20 repair gamma ray PLD in the presence or absence of 3AB or  
NU1025 as follows. L1210 cells were maintained in culture  
until they had attained stationary phase (>10<sup>6</sup>cells/ml). They  
were diluted to 1.5 x 10<sup>5</sup>/ml in conditioned medium from  
stationary-phase cultures to prevent further cell  
25 division. Replicate 2ml samples of cells in plastic bijoux  
were held on ice prior to and immediately following 8 Gray  
gamma ray irradiation. 1ml of 3x final concentration of  
compounds 3AB or NU1025 made up in conditioned medium from  
stationary cultures was added (to give final concentrations  
30 of 10<sup>6</sup>cells/ml in 1% DMSO ± 10mM 3AB or 200 $\mu$ M NU1025) and  
the cells were incubated at 37°C for 0, 2 or 4 hours prior  
to resuspending in drug-free medium and seeding for colony  
formation. Unirradiated stationary phase cultures  
incubated at 37°C for 0, 2 or 4 hours with 1% DMSO ± 10mM  
35 3AB or 200 $\mu$ M NU1025 were used as appropriate controls for  
determining relative cell survival. In the absence of PARP  
inhibitor cell survival increased with time allowed for PLD  
repair to take place. For example, when seeded immediately  
after irradiation (no repair) only about 0.2% of the cells  
survived, but after a 4 hour repair period this had

increased to 0.7%. It was observed that both 3AB and NU1025 blocked this repair.

The cytotoxic effects of the PARP inhibiting compounds alone has also been investigated. In one set of experiments, the LD<sub>50</sub> values for a 24 hours exposure of L1210 cells were 14 ± 1.0mM (3AB); 6.0 ± 1.5mM (BZ) and 1.6 ± 0.1mM (NU1025). The LD<sub>50</sub> values differed by <3-fold from the IC<sub>50</sub> values but maintained the same rank order with respect to their potency as PARP inhibitors. In agreement with the growth inhibition data there was >10-fold difference between the concentration of NU1025 needed to produce maximal potentiation of TM cytotoxicity and the concentration needed to produce cytotoxicity per se.

15

In respect of DNA strand break assays carried out, it was found that a 1 hour treatment with TM resulted in a concentration-dependent increase in the rate of elution which provides a measure of the extent of DNA strand breakage. Changes in DNA strand break levels were detectable at levels of TM as low as 150µM, a concentration which reduced survival by about 30%. All the compounds were tested for their ability to produce strand breaks when used alone. A 24 hour incubation of cells with 1mM NU1025, and 20mM 3AB or BZ had no significant effect on DNA strand break levels compared to untreated cells. However, coincubation for 1 hour of a fixed concentration of TM (150µM) with increasing concentrations of all PARP inhibitors tested caused a progressive increase in the rate of elution (extent of strand breakage) compared to TM alone.

The results for all the 3 representative compounds mentioned have been summarised by plotting values of a parameter related to extent of strand breakage versus inhibitor concentration. For all the compounds, the strand breakage increased linearly with increasing concentration, but values started increasing significantly for NU1025 at about 100µM, whereas concentrations above 3mM and 5mM were

required to significantly increase values for BZ and 3AB respectively. Again, the rank order and potency of the compounds in the DNA strand break assay demonstrated an excellent correlation with *in vitro* PARP inhibitory potency.

Overall, it is believed that the studies carried out give clear evidence that the PARP inhibitory characteristics of compounds of this invention reflects an ability 10 of these compounds to potentiate the cytotoxicity of DNA damaging agents such as certain cytotoxic antitumour drugs and radiation used in radiotherapy. Accordingly, such compounds should be especially useful for administration in conjunction with such cytotoxic drugs or radiotherapy to 15 potentiate their effect in the course of medical treatment as hereinbefore indicated.

20 Summary

Although the present invention should be regarded overall as comprising each and every novel feature or combination of features disclosed herein, the main aspects 25 of the invention comprise, principally but not exclusively, broadly the following:-

- (i) Novel compounds of formula (I), (II) or (IV) as defined herein;
- (ii) Compounds of formula (I), (II) or (IV) with substituents as hereinbefore defined (including salts thereof) for therapy or for use in medicine and in the manufacture of medical preparations, useful for example as PARP inhibitors to be administered in conjunction with cytotoxic drugs or with radiotherapy to potentiate the effectiveness of the latter in treatment of cancer;
- (iii) Processes for the preparation of novel compounds of formula (I), (II) or (IV) as defined herein, including any novel intermediate compounds produced in carrying out such processes;

- (iv) Pharmaceutical formulations comprising a compound of formula (I), (II) or (IV) as defined herein together with a pharmaceutically acceptable carrier therein;
- 5 (v) Processes for the preparation of a pharmaceutical formulation as defined in (iv) above, e.g. by methods referred to herein;
- 10 (vi) Quinazolinone compounds of formula (II),- possibly representing molecularly rearranged compounds of formula (IV), as herein disclosed, for therapy or for use in medicine and in the manufacture of medical preparations, useful for example as PARP inhibitors to be administered in conjunction with cytotoxic drugs or with radiotherapy to potentiate the effectiveness of the latter in treatment of cancer,
- 15 and pharmaceutical formulations comprising said quinazolinone compounds.

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TABLE I

INHIBITOR	IC <sub>50</sub> (mM) ± SE INHIBITOR ALONE	IC <sub>50</sub> (mM) ± SE INHIBITOR + 100μM TM
3-AMINOBENZAMIDE	6.7 ± 0.2	2.5 ± 0.1
BENZAMIDE	2.5 ± 0.3	0.84 ± 0.12
NU1025	0.41 ± 0.06	0.04 ± 0.003

TABLE II

INHIBITOR	CONCENTRATION	DEF <sub>10</sub>
3-AMINOBENZAMIDE	1mM	2.4 ± 0.3
	5mM	4.1 ± 0.4
BENZAMIDE	1mM	4.0 ± 0.7
	3mM	6.9 ± 0.2
NU1025	10μM	2.0 ± 0.2
	50μM	4.0 ± 0.5
	100μM	5.1 ± 0.7

TABLE III

House Number	Name	Structure	% Inhibition		
			10µM	30µM	100µM
Ref	3-hydroxybenzamide  C <sub>7</sub> H <sub>7</sub> NO <sub>2</sub>  MW = 137		35	59	81
Ref	3-methoxybenzamide  C <sub>8</sub> H <sub>9</sub> NO <sub>2</sub>  MW = 151		55	78	89
Ref	3-aminobenzamide  C <sub>7</sub> H <sub>8</sub> N <sub>2</sub> O <sub>2</sub>  MW = 136		36	63	79
NU1005	3-benzyloxybenzamide  C <sub>14</sub> H <sub>13</sub> NO <sub>2</sub>  MW = 227		34	60 59	76
NU1006	3-(4-methoxybenzyloxy)benzamide  C <sub>15</sub> H <sub>15</sub> NO <sub>3</sub>  MW = 257		58	74	insol.
NU1007	3-(4-nitrobenzyloxy)benzamide  C <sub>14</sub> H <sub>12</sub> N <sub>2</sub> O <sub>4</sub>  MW = 272		26	insol.	insol.

TABLE III (contd.)

House Number	Name	Structure	% Inhibition		
			10µM	30µM	100µM
NU1008	3-(cyclohexylmethoxy) benzamide  C <sub>14</sub> H <sub>19</sub> NO <sub>2</sub>  MW = 233		insol.	insol.	insol.
NU1013	3-(4-azidobenzoyloxy) benzamide  C <sub>14</sub> H <sub>12</sub> N <sub>4</sub> O <sub>2</sub>  MW = 268		49	28	43
NU1014	3-(4-bromobenzoyloxy) benzamide  C <sub>14</sub> H <sub>12</sub> BrNO <sub>2</sub>  MW = 306		23	36	insol.
NU1015	3-(4-fluorobenzoyloxy) benzamide  C <sub>14</sub> H <sub>12</sub> FNO <sub>2</sub>  MW = 245		20	52 51	insol.
NU1016	3-(4-aminobenzoyloxy) benzamide  C <sub>14</sub> H <sub>14</sub> N <sub>2</sub> O <sub>2</sub>  MW = 242		40	66	84
NU1017	3-(3-nitrobenzoyloxy) benzamide  C <sub>14</sub> H <sub>12</sub> N <sub>2</sub> O <sub>4</sub>  MW = 272		21	40	insol. insol.

TABLE III (contd.)

House Number	Name	Structure	% Inhibition		
			10µM	30µM	100µM
NU1019	3-(S-bromopentyloxy) benzamide  C <sub>12</sub> H <sub>16</sub> BrNO <sub>2</sub>  MW = 268			54	81
NU1020	3-(piperonyloxy) benzamide  C <sub>15</sub> H <sub>13</sub> NO <sub>4</sub>  MW = 271		51	70	93
NU1022	3-(8-adenos-9-yloctyloxy) benzamide  C <sub>20</sub> H <sub>26</sub> N <sub>6</sub> O <sub>2</sub>  MW = 382		30		
NU1023	3-[5-(6-chloropurin-9-yl) pentyloxy]benzamide  C <sub>17</sub> H <sub>18</sub> CIN <sub>5</sub> O <sub>2</sub>  MW = 360		26	45	71
NU1024	3-(5-adenos-9- ylpentyloxy)benzamide  C <sub>17</sub> H <sub>20</sub> N <sub>6</sub> O <sub>2</sub>  MW = 340		16	42	67
NU1025	8-hydroxy-2-methyl- quinazolin-4-[3H]one  C <sub>9</sub> H <sub>8</sub> N <sub>2</sub> O <sub>2</sub>  MW = 176		92 1µM = 63 0.1µM = 18 0.5µM = 59 1.0µM = 68  IC <sub>50</sub> = 0.4 µM	92	96.

TABLE III (contd.)

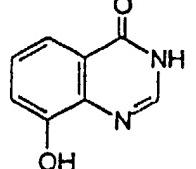
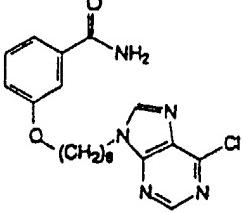
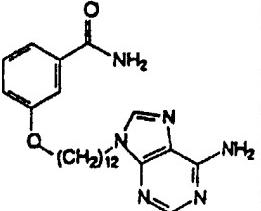
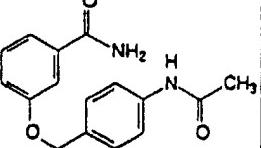
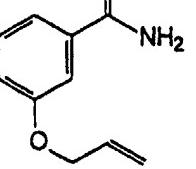
House Number	Name	Structure	% Inhibition		
			10µM	30µM	100µM
NU1026	8-hydroxyquinazolin-4-[3H]one C <sub>8</sub> H <sub>6</sub> N <sub>2</sub> O <sub>2</sub> MW = 162		78 0.5µM = 18 1.0µM = 38 2.0µM = 54 IC <sub>50</sub> = 2 µM	87	95
NU1027	3-[8-(6-chloropurin-9-yl)octyloxy]benzamide C <sub>20</sub> H <sub>24</sub> CIN <sub>5</sub> O <sub>2</sub> MW = 402		30	insol.	insol.
NU1029	3-(12-adenos-9-yl)dodecyloxybenzamide C <sub>24</sub> H <sub>34</sub> N <sub>6</sub> O <sub>2</sub> MW = 438		26	51	74
NU1030	3-(N-acetyl-4-aminobenzyloxy)benzamide C <sub>16</sub> H <sub>16</sub> N <sub>2</sub> O <sub>3</sub> MW = 284		49	70	84
NU1031	3-allyloxybenzamide C <sub>10</sub> H <sub>11</sub> NO <sub>2</sub> MW = 177		21	58	80

TABLE III (contd.)

House Number	Name	Structure	% Inhibition		
			10µM	30µM	100µM
NU1034	2,3-methylenedioxy benzamide  C <sub>8</sub> H <sub>7</sub> NO <sub>3</sub>  MW = 165		66  IC <sub>50</sub> = 5.3µM	-	94
NU1036	3-(4-trifluoromethyl benzyloxy)benzamide  C <sub>15</sub> H <sub>12</sub> F <sub>3</sub> NO <sub>2</sub>  MW = 295		3	13	10
NU1037	3-(4-cyanobenzyloxy) benzamide  C <sub>15</sub> H <sub>12</sub> N <sub>2</sub> O <sub>2</sub>  MW = 252		33	insol.	insol.
NU1038	3-butoxybenzamide  C <sub>11</sub> H <sub>15</sub> NO <sub>2</sub>  MW = 193		23	50	78
NU1039	3-pentyloxybenzamide  C <sub>12</sub> H <sub>17</sub> NO <sub>2</sub>  MW = 207		18	45	66
NU1040	2- <i>t</i> -butylbenzoxazole-4-carboxamide  C <sub>12</sub> H <sub>14</sub> N <sub>2</sub> O <sub>2</sub>  MW = 218		87  IC <sub>50</sub> = 8.4µM	88	93

TABLE III (contd.)

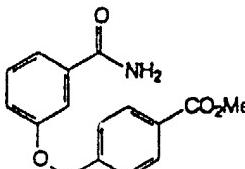
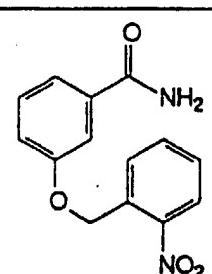
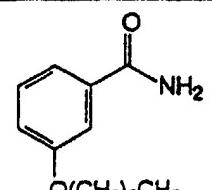
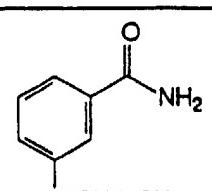
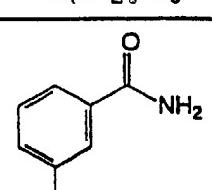
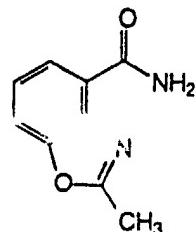
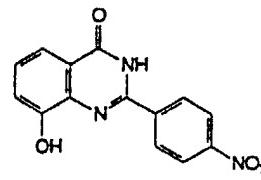
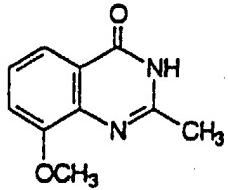
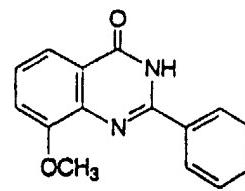
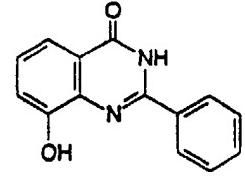
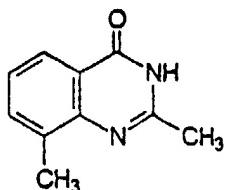
House Number	Name	Structure	% Inhibition		
			10µM	30µM	100µM
NU1041	3-(4-carboxymethylbenzyloxy)benzamide  C <sub>16</sub> H <sub>15</sub> NO <sub>4</sub>  MW = 285		insol.	insol.	insol.
NU1042	3-(2-nitrobenzyloxy)benzamide  C <sub>14</sub> H <sub>12</sub> N <sub>2</sub> O <sub>4</sub>  MW = 272		insol.	insol.	insol.
NU1043	3-hexyloxybenzamide  C <sub>13</sub> H <sub>19</sub> NO <sub>2</sub>  MW = 221		15	39	insol.
NU1044	3-heptyloxybenzamide  C <sub>14</sub> H <sub>21</sub> NO <sub>2</sub>  MW = 235		26	insol.	insol.
NU1045	3-octyloxybenzamide  C <sub>15</sub> H <sub>23</sub> NO <sub>2</sub>  MW = 249		insol.	insol.	insol.

TABLE III (contd.)

House Number	Name	Structure	% Inhibition		
			10μM	30μM	100μM
NU1048	3-phenethoxybenzamide  C <sub>15</sub> H <sub>15</sub> NO <sub>2</sub> MW = 241		30	50	72
NU1050	3-cinnamylxybenzamide  C <sub>16</sub> H <sub>15</sub> NO <sub>2</sub> MW = 253		23	insol.	insol.
NU1051	2-phenylbenzoxazole-4-carboxamide  C <sub>14</sub> H <sub>10</sub> N <sub>2</sub> O <sub>2</sub> MW = 238		82  IC <sub>50</sub> = 2.1 μM		
NU1052	3-(4-carboxybenzyloxy)benzamide  C <sub>15</sub> H <sub>13</sub> NO <sub>4</sub> MW = 271				
NU1054	2-(4-methoxyphenyl)benzoxazole-4-carboxamide  C <sub>15</sub> H <sub>12</sub> N <sub>2</sub> O <sub>2</sub> MW = 252			IC <sub>50</sub> = 1.1 μM	

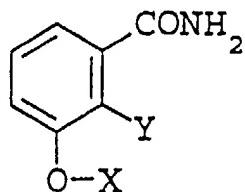
TABLE III (contd.)

House Number	Name	Structure	% Inhibition		
			10µM	30µM	100µM
NU1056	2-methylbenzoxazole-4-carboxamide  C <sub>9</sub> H <sub>8</sub> N <sub>2</sub> O <sub>2</sub>  MW = 176		IC <sub>50</sub> = 9.5µM		
NU1057	8-hydroxy-2-(4-nitrophenyl)-quinazolin-4-one  C <sub>14</sub> H <sub>9</sub> N <sub>3</sub> O <sub>4</sub>  MW = 283.2		92  IC <sub>50</sub> = 0.23µM		
NU1063	8-methoxy-2-methylquinazolin-4[3H]-one  C <sub>10</sub> H <sub>10</sub> N <sub>2</sub> O <sub>2</sub>  MW = 190.2		IC <sub>50</sub> = 0.78µM		
NU1065	8-methoxy-2-phenylquinazolin-4[3H]-one  C <sub>15</sub> H <sub>12</sub> N <sub>2</sub> O <sub>2</sub>  MW = 252.27		IC <sub>50</sub> = 4.2µM		
NU1068	8-hydroxy-2-phenylquinazolin-4[3H]-one  C <sub>14</sub> H <sub>10</sub> N <sub>2</sub> O <sub>2</sub>  238.24		IC <sub>50</sub> = 0.53µM		
NU1069	2,8-dimethylquinazolin-4[3H]-one  C <sub>10</sub> H <sub>10</sub> N <sub>2</sub> O <sub>2</sub>  174.2		IC <sub>50</sub> = 0.2µM		

CLAIMS

1. A compound selected from: (A), a 3-substituted benzamide compound having the general structural formula I

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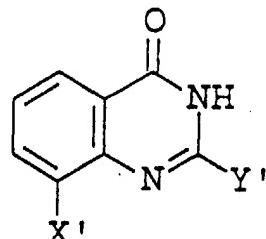


I

10

or a pharmaceutically acceptable salt thereof, and  
(B), a quinazolinone compound having the general structural formula II

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II

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or a pharmaceutically acceptable salt thereof,

characterised in that in structural formula I

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(i) Y is hydrogen, and

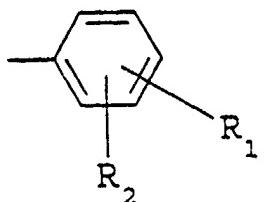
X is -CH<sub>2</sub>-Z

wherein

30

Z represents alkyl, an optionally substituted aralkyl group, -CH=CHR (where R is H, alkyl or an optionally substituted phenyl group), cyclohexyl, or a group having the structural formula III

35



III

where R<sub>1</sub> is selected from H, alkoxy,

5

$\text{NO}_2$ ,  $\text{N}_3$ ,  $\text{NH}_2$ ,  $\text{NHCOR}_3$  ( $R_3$  being alkyl or aryl),  $\text{CO}_2\text{R}_4$  ( $R_4$  being H or alkyl), alkyl, hydroxyalkyl,  $\text{CW}_3$  or W (W being halide), and CN,

and where

$R_2$  is H,

10

or where  $R_1$  and  $R_2$  together represent a group  $-\text{O}-\text{CHR}_5-\text{O}-$  bridging adjacent ring C's with  $R_5$  being H, alkyl or an optionally substituted aralkyl or aryl group

or

15

(ii) Y is hydrogen, and

X is  $-(\text{CH}_2)_n-\text{Z}$

wherein n is in the range of 5 to 12, and

Z is halide or a purin-9-yl moiety;

20 or

25

(iii) Y and X together form a bridge  $-\text{Y}-\text{X}-$  that represents the grouping



where  $R_5$  is as specified above,

25

and in structural formula II

30

X' represents hydroxyl, alkyl, alkoxy or an optionally substituted aryl (e.g. phenyl) or aralkyl (e.g. benzyl) group,

and

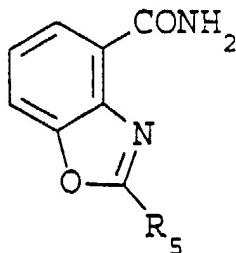
35

Y' represents hydrogen, alkyl or an optionally substituted aryl (e.g. phenyl) or aralkyl (e.g. benzyl) group.

2. A compound as claimed in Claim 1 wherein the or each alkyl group present, either as such or as a moiety in an alkoxy or other group, within the group Z and/or the group R<sub>5</sub> when the compound has the structural formula I, or 5 within the groups X' and Y' when the compound conforms to structural formula II, contains 1-6 carbon atoms.

3. A compound as claimed in Claim 1 which is a benzoxazole-4-carboxamide compound having a structural 10 formula IV

15

**IV**

4. A compound as claimed in Claim 3 wherein R<sub>5</sub> is 20 selected from alkyl, phenyl, naphthyl and pyridyl.

5. A compound as claimed in Claim 3 wherein R<sub>5</sub> is a C<sub>1-6</sub> alkyl group.

25 6. A compound as claimed in Claim 3 which is one of the following:

- (a) 2-methylbenzoxazole-4-carboxamide;
- (b) 2-t-butylbenzoxazole-4-carboxamide;

30 (c) 2-phenylbenzoxazole-4-carboxamide;  
 (d) 2-(4-methoxyphenyl)benzoxazole-4-carboxamide;  
 or a pharmaceutically acceptable salt of any of the above compounds (a) to (d).

35 7. A compound as claimed in Claim 1 or 2 having the general formula I wherein X is a benzyl or substituted benzyl group.

8. A compound as claimed in Claim 1 or 2 having the general formula I wherein X is a benzyl group having a

substituent selected from 2-NO<sub>2</sub>, 4-CH<sub>3</sub>, 4-CO<sub>2</sub>H, 4-CO<sub>2</sub>CH<sub>3</sub>, 4-CONH<sub>2</sub>, 4-CN, 4-CH<sub>2</sub>OH and 4-NHCOPh.

9. A compound as claimed in Claim 1 which is one of the  
5 following:

- (a) 3-benzylbenzamide;
- (b) 3-(4-methoxybenzyl)benzamide;
- (c) 3-(4-nitrobenzyl)benzamide;
- (d) 3-(4-azidobenzoxy)benzamide;
- 10 (e) 3-(4-bromobenzyl)benzamide;
- (f) 3-(4-fluorobenzyl)benzamide;
- (g) 3-(4-aminobenzyl)benzamide;
- (h) 3-(3-nitrobenzyl)benzamide;
- (i) 3-(3,4-methylenedioxyphenylmethoxy)benzamide;  
15 or 3-(piperonyloxy)benzamide;
- (j) 3-(N-acetyl-4-aminobenzyl)benzamide;
- (k) 3-(4-trifluoromethylbenzyl)benzamide;
- (l) 3-(4-cyanobenzyl)benzamide;
- (m) 3-(4-carboxymethylbenzyl)benzamide;
- 20 (n) 3-(2-nitrobenzyl)benzamide;
- (o) 3-(4-carboxybenzyl)benzamide;

or a pharmaceutically acceptable salt any one of the above compounds (a) to (o).

25 10. A compound as claimed in Claim 1 which is one of the following:

- (a) 3-(5-bromopentyloxy)benzamide,
- (b) 3-(8-adenos-9-yloctyloxy)benzamide,
- (c) 3-[5-(6-chloropurin-9-yl)pentyloxy]benzamide,  
30 (d) 3-(5-adenos-9-ylpentyl)benzamide,
- (e) 3-[8-(6-chloropurin-9-yl)octyloxy]benzamide,
- (f) 3-[12-(6-chloropurin-9-yl)dodecyloxy]benzamide,
- (g) 3-(12-adenos-9-yl)dodecyloxy)benzamide.

35 or a pharmaceutically acceptable salt of any of the above compounds (a) to (g).

11. A compound as claimed in Claim 1 which is a 3-allyloxybenzamide, a 3-cinnamyloxybenzamide, or a pharmaceuti-

cally acceptable salt thereof.

12. A compound as claimed in Claim 1 which is one of the following:

- 5 (a) 3-butoxybenzamide,  
(b) 3-pentyloxybenzamide,  
(c) 3-hexyloxybenzamide,  
(d) 3-heptyloxybenzamide,  
(e) 3-octyloxybenzamide,

10 or a pharmaceutically acceptable salt thereof.

13. A compound as claimed in Claim 1 or 2 which is a quinazolinone compound having the general structural formula II wherein Y' is phenyl or a phenyl group having a  
15 substituent selected from -NO<sub>2</sub>, -NH<sub>2</sub>, -OH and alkyl.

14. A compound as claimed in Claim 13 wherein X' is hydroxyl.

20 15. A compound as claimed in Claim 1 or 2 which is a quinazolinone compound having the general structural formula II wherein X' is hydroxyl and Y' is alkyl.

25 16. A compound as claimed in any of Claims 13 to 15 produced by a molecular rearrangement of a benzoxazole-4-carboxamide compound.

30 17. A quinazolone derivative produced by a molecular rearrangement of a benzoxazole-4-carboxamide compound claimed in Claim 3.

18. A quinazolinone compound as claimed in Claim 1 which is one of the following:

- 35 (a) 8-hydroxy-2-methylquinazolin-4-[3H]one;  
(b) 8-hydroxyquinazolin-4-[3H]one;  
(c) 8-hydroxy-2-(4-nitrophenyl)-quinazolin-4-one;  
(d) 8-methoxy-2-methylquinazolin-4[3H]-one;  
(e) 8-methoxy-2-phenylquinazolin-4[3H]-one;  
(f) 8-hydroxy-2-phenylquinazolin-4[3H]-one;

(g) 2,8-dimethylquinazolin-4[3H]-one.

19. A process for preparing a compound as claimed in Claim 1 comprising a step of reacting a 3-hydroxybenzamide compound with an alkylating agent under base-catalysed reaction conditions.

20. A compound as claimed in any one of Claims 1 to 18 or use in therapy as an active PARP-inhibiting substance.

10

21. A pharmaceutical formulation or composition containing a compound as claimed in Claim 20 in unit dosage form made up for administration to a mammal in need of treatment with a PARP-inhibiting agent.

15

22. A pharmaceutical formulation or composition for medical use comprising an effective PARP-inhibiting amount of a compound as claimed in any one of Claims 1 to 18.

20

23. A pharmaceutical formulation or composition as claimed in Claim 21 or 22 for use in conjunction with cytotoxic agents in antitumour therapy.

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24. Use of a compound as claimed in any one of Claims 1 to 18 for the manufacture of a medical or veterinary preparation for inhibiting PARP activity and increasing cytotoxic effectiveness of cytotoxic drugs or radiotherapy administered in conjunction therewith for the treatment of tumours in mammals.

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**INTERNATIONAL SEARCH REPORT**

Int. Appl. No.  
PCT/GB 95/00513

A. CLASSIFICATION OF SUBJECT MATTER					
IPC 6	C07C235/46	C07D263/54	C07D239/90	C07D239/91	C07D317/46
	C07D327/04	A61K31/41	A61K31/47	A61K31/335	A61K31/39

According to International Patent Classification (IPC) or to both national classification and IPC

B. FIELDS SEARCHED

Minimum documentation searched (classification system followed by classification symbols)

IPC 6 C07C C07D A61K

Documentation searched other than minimum documentation to the extent that such documents are included in the fields searched

Electronic data base consulted during the international search (name of data base and, where practical, search terms used)

C. DOCUMENTS CONSIDERED TO BE RELEVANT

Category *	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
X	EP,A,0 411 766 (MERCK AND CO., INC; USA) 6 February 1991 see example 7 ---	1,2,17
X	US,A,3 169 129 (AMERICAN CYANAMID CO.) 9 February 1965 & US-RE-26565 (29 April 1969) see examples 6,15 ---	1,2,17
X	US,A,4 281 127 (HOFFMANN-LA ROCHE INC.) 28 July 1981 see example 28 ---	1,2,17

Further documents are listed in the continuation of box C.

Patent family members are listed in annex.

\* Special categories of cited documents :

- \*A\* document defining the general state of the art which is not considered to be of particular relevance
- \*B\* earlier document but published on or after the international filing date
- \*L\* document which may throw doubts on priority claim(s) or which is cited to establish the publication date of another citation or other special reason (as specified)
- \*O\* document referring to an oral disclosure, use, exhibition or other means
- \*P\* document published prior to the international filing date but later than the priority date claimed

\*T\* later document published after the international filing date or priority date and not in conflict with the application but cited to understand the principle or theory underlying the invention

\*X\* document of particular relevance; the claimed invention cannot be considered novel or cannot be considered to involve an inventive step when the document is taken alone

\*Y\* document of particular relevance; the claimed invention cannot be considered to involve an inventive step when the document is combined with one or more other such documents, such combination being obvious to a person skilled in the art.

\*&\* document member of the same patent family

Date of the actual completion of the international search	Date of mailing of the international search report
11 July 1995	01.08.95
Name and mailing address of the ISA	Authorized officer
European Patent Office, P.B. 5818 Patentlaan 2 NL - 2280 HV Rijswijk Tel. (+ 31-70) 340-2040, Tx. 31 651 epo nl, Fax (+ 31-70) 340-3016	Slootweg, A

## INTERNATIONAL SEARCH REPORT

International Application No  
PCT/GB 95/00513

## C.(Continuation) DOCUMENTS CONSIDERED TO BE RELEVANT

Category *	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
X	J. MED. CHEM., vol. 22, no. 11, 1979 pages 1354-1357, MITSCHER ET AL. 'Quinolone antimicrobial agents. 2. Methylenedioxy positional isomers of oxolinic acid' Compound 12 (p.1355) see page 1366, column 1, paragraph 5 ---	1,2
X	J. CHEM. SOC., 1962 PART II, pages 2549-2556, PARTRIDGE ET AL. 'Cyclic amidines. Part XV. Derivatives of tricycloquinazoline.' see page 2549, compound (II)b ---	1,2,13, 16,17
X	CHEMICAL ABSTRACTS, vol. 54, no. 13, 10 July 1960 Columbus, Ohio, US; abstract no. 13054e, L.V.COATES ET AL. 'Preparation and evaluation of some phenolic ethers as antifungal agents' see abstract & J. PHARM. AND PHARMACOL., vol. 11, 1959 pages 240T-249T, ---	1,2,12, 20-22
X	CHIM. THER., vol. 2, no. (4), 1967 page 231-9 MAILLARD ET AL. 'Dérivés de la (3H) quinazoline-4 doués de la propriétés anti-inflammatoires II. - Dérivés substitués dans le noyau aromatique et produits voisins.' see page 232; table I ---	1,2,17, 18,20-23
X	DE,A,41 42 366 (DR. KARL THOMAE GMBH) 24 June 1993 description and claims see example 35 ---	1,2,16, 17
X	Z. NATURFORSCH., vol. 133C, no. (7-8), 1978 C: BIOSC., page 459-64 DALLACKER ET AL. 'Über Amino-benzo[1,3]dioxole.' see page 460; example 2A ---	1
		-/-

## INTERNATIONAL SEARCH REPORT

Int'l Application No  
PCT/GB 95/00513

## C.(Continuation) DOCUMENTS CONSIDERED TO BE RELEVANT

Category *	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
X	INDIAN J. CHEM., vol. 24B, no. (11), 1985 SECT.B (IJSBDB, 03764699), page 1182-4 SINHA ET AL. 'Quinazolones: part XI - effect of substituents on Claisen rearrangement of allyloxyquinazolones.' see page 1183, column 2, last paragraph ---	1,2,17, 18
X	INDIAN J. CHEM., vol. 16B, no. (12), 1978 page 1067-72 JOSHI ET AL. 'Studies in qu...azolones: Part I - Synthesis & spectral characteristics of substituted 2-isopropyl-4(3H)-quinazolones.' see page 1068; examples XVII, XIX,XXI ---	1,2,17
X	DE,A,32 20 898 (DR. KARL THOMAE GMBH) 8 December 1983 see page 14, line 16-18 and page 11, line 10-12 ---	1,2,16, 17,20-23
X	EP,A,0 054 132 (DR. KARL THOMAE GMBH) 23 June 1982 see examples 5,30 ---	1,2,13, 16,17, 20-23
X	WO,A,93 12095 (PFIZER LTD.) 24 June 1993 see examples 1,2,5-10,16,17,22 and Preparations 4,8,10,11,15,21 ---	1,2,16, 17,20-24
X	MH. CHEM., vol. 92, 1961 pages 1147-1154, SPITELLER. G. 'Der o-Effekt in den Massenspektren aromatischer Verbindungen' see page 1151 ---	1,2
X	AGRIC. BIOL. CHEM., vol. 44 (2), 1980 pages 235-243, TOSHIKAZU ET AL. 'Synthesis of furoquinolines' see p. 243, Compound XIIe ---	1,2
X	J. HETEROC. CHEM., vol. 19 (1), 1982 pages 171-176, DEAN ET AL. 'N-ethoxycarbonylamides as starting materials and intermediates in the synthesis of heterocyclic compounds' see p.172, compounds 6e and 6m ---	1,2,13
		-/-

## INTERNATIONAL SEARCH REPORT

International Application No  
PCT/GB 95/00513

## C.(Continuation) DOCUMENTS CONSIDERED TO BE RELEVANT

Category *	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
X	J.CHEM.SOC. (B), vol. 67,no. (5), 1967 pages 449-454, ARMAREGO ET AL. 'Quinazolines. Part IX. Covalent hydration in neutral species of substituted quinazolines.' see page 450; table 1 ---	1,2
Y	WO,A,91 18591 (COLLINS,FARZANEH,SHALL,TAVASSOLI) 12 December 1991 see the whole document ---	1,2,7,8, 12,19-24
Y	EP,A,0 305 008 (OXI-GENE, INC.) 1 March 1989 see the whole document ---	1,2,7,8, 12,19-24
A	WO,A,93 07868 (OCTAMER INC.) 29 April 1993 see the whole document -----	1-24

**INTERNATIONAL SEARCH REPORT**

Information on patent family members

Int	lational Application No
PCT/GB 95/00513	

Patent document cited in search report	Publication date	Patent family member(s)		Publication date
EP-A-411766	06-02-91	AU-B-	626699	06-08-92
		AU-A-	5860790	03-01-91
		CA-A-	2020073	04-01-91
		DE-D-	69006131	03-03-94
		DE-T-	69006131	21-07-94
		IL-A-	94915	07-10-94
		JP-A-	3115271	16-05-91
		US-A-	5240928	31-08-93
US-A-3169129	09-02-65	NONE		
US-A-1781127	28-07-81	AT-T-	7496	15-06-84
		AU-O-	521014	11-03-82
		AU-A-	6024480	12-03-81
		CA-A-	1134825	02-11-82
		EP-A,B	0023594	11-02-81
		JP-C-	1364987	09-02-87
		JP-A-	56025167	10-03-81
		JP-B-	61029346	05-07-86
DE-A-4142366	24-06-93	DE-A-	4123341	21-01-93
		EP-A-	0529253	03-03-93
		JP-A-	5247074	24-09-93
DE-A-3220898	08-12-83	CA-A-	1217768	07-02-87
EP-A-0054132	23-06-82	DE-A-	3046871	22-07-82
		DE-A-	3115447	30-12-82
		AT-T-	9797	15-10-84
		AU-B-	543568	26-04-85
		AU-A-	7847381	17-06-82
		CA-A-	1217767	07-02-87
		GB-A,B	2090249	07-07-82
		JP-A-	57123170	31-07-82
		SU-A-	1056900	23-11-83
		US-A-	4379788	12-04-83
WO-A-9312095	24-06-93	CA-A-	2122360	24-06-93
		EP-A-	0628032	14-12-94
		FI-A-	942769	10-06-94

## INTERNATIONAL SEARCH REPORT

Information on patent family members

International Application No  
PCT/GB 95/00513

Patent document cited in search report	Publication date	Patent family member(s)		Publication date
WO-A-9312095		JP-T-	7502029	02-03-95
WO-A-9118591	12-12-91	AU-B-	645812	27-01-94
		AU-A-	7880491	31-12-91
		EP-A-	0531370	17-03-93
		GB-A,B	2244646	11-12-91
EP-A-0305008	01-03-89	AT-T-	106749	15-06-94
		AU-A-	2150888	02-03-89
		DE-D-	3850007	14-07-94
		DE-T-	3850007	22-09-94
		ES-T-	2056902	16-10-94
		IL-A-	87525	30-05-94
		JP-A-	1096122	14-04-89
		US-A-	5340565	23-08-94
WO-A-9307868	29-04-93	AU-A-	1260992	21-05-93
		EP-A-	0609211	10-08-94
		JP-T-	7502975	30-03-95